

# Trans,trans,trans-Diethanoldiquinaldinatoiron(II)

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***trans,trans,trans*-Diethanoldiquinaldinatoiron(II)****Kunitoyo Osawa *et al.*****Synopsis****Queries and comments**

Please supply or correct as appropriate all **bold underlined** text.

**Subject index**

The following terms will be used to index your paper. Authors wishing to recommend additional index entries should give these below.

*trans,trans,trans*-Diethanoldiquinaldinatoiron(II)

**Inorganic formula index**

Note that, for coordination complexes, the ligands are listed in alphabetic order. This means that the indexing term may differ from the IUPAC formula used elsewhere in the paper.

[Fe(C<sub>10</sub>H<sub>6</sub>NO<sub>2</sub>)<sub>2</sub>(C<sub>2</sub>H<sub>6</sub>O)<sub>2</sub>]

**Organic formula index**

All residues containing organic carbon are included in this index.

C<sub>24</sub>H<sub>24</sub>FeN<sub>2</sub>O<sub>6</sub>

**Author index**

Authors' names will normally be arranged alphabetically under their family name and this is commonly their last name. Prefixes (*van, de etc.*) will only be taken into account in the alphabetization if they begin with a capital letter. Authors wishing their names to be alphabetized differently should indicate this below.

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*trans,trans,trans*-Diethanoldiquinaldinatoiron(II)

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## Key indicators

Single-crystal X-ray study  
 $T = 123\text{ K}$   
 Mean  $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$   
 $R$  factor = 0.039  
 $wR$  factor = 0.069  
 Data-to-parameter ratio = 16.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title complex, *trans,trans,trans*-[Fe<sup>II</sup>(C<sub>10</sub>H<sub>6</sub>NO<sub>2</sub>)<sub>2</sub>-(C<sub>2</sub>H<sub>6</sub>O)<sub>2</sub>], is centrosymmetric and the quinaldinate ligands form five-membered chelate rings. The geometry of the complex is distorted octahedral, with a *trans*-FeN<sub>2</sub>O<sub>4</sub> chromophore. The hydroxy H atom forms an intermolecular hydrogen bond with the carbonyl O atom of the quinaldinate ligand.

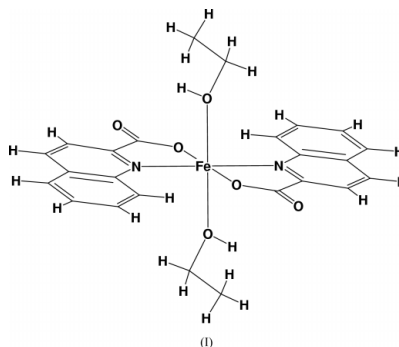
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## Comment

Quinaldic acid is associated with tryptophan metabolism (Zhou *et al.*, 1989) and is used as a reagent for solvent extraction of divalent transition metal ions (Högberg *et al.*, 1985). There are few structural studies of quinaldinate complexes in spite of numerous studies of related picolinato complexes. Only the Cu<sup>2+</sup> (Haendler, 1986), Rh<sup>+</sup> (Lamprecht *et al.*, 1986) and Ga<sup>3+</sup> (Li *et al.*, 1996) complexes have been structurally characterized. Therefore, structural information of another transition metal complex is desired.



The title complex, (I), is monomeric and has a distorted octahedral structure, with the central atom lying on an inversion center (Fig. 1 and Table 1). The complex has a *trans,trans,trans*-geometry with respect to three kinds of donors. The quinaldinate acts as a planar *N,O*-bidentate ligand and forms a five-membered chelate ring upon coordination. Two quinaldinate ligands are connected by weak intramolecular hydrogen bonds; the distance between atoms C9 and O1<sup>i</sup> is 3.152 (3) Å [symmetry code: (i)  $-x, 1 - y, -z$ ].

There exists a strong hydrogen bond between an ethanol molecule and the uncoordinated O atom of a neighboring quinaldinate ligand. The distance between atoms O3 and O2<sup>ii</sup> is 2.694 (3) Å [symmetry code: (ii)  $1 - x, 1 - y, -z$ ]. The hydrogen bonds form one-dimensional molecular chains parallel to the *a* axis. The chains are connected by weak hydrogen bonds (Table 2).

Experimental

The title complex was prepared under an N<sub>2</sub> atmosphere using Schlenk techniques. To a solution of Fe(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O (0.134 g, 0.397 mmol) in 1.6 ml ethanol was added a solution containing quinaldic acid (0.173 g, 0.999 mmol) in ethanol (6 ml) and triethylamine (140 ml, 0.100 mmol). After vigorous stirring, the solution was allowed to stand for 2 d to afford red-violet crystals suitable for X-ray analysis. The IR spectrum shows a ν(CO<sub>2</sub>) band at 1628 cm<sup>-1</sup>. The electronic spectrum in DMF exhibits an absorption maximum at 527 nm (ε = 795).

Crystal data

[Fe(C<sub>10</sub>H<sub>6</sub>NO<sub>2</sub>)<sub>2</sub>(C<sub>2</sub>H<sub>6</sub>O)<sub>2</sub>]  
 M<sub>r</sub> = 492.30  
 Monoclinic, P2<sub>1</sub>/n  
 a = 5.816 (2) Å  
 b = 9.557 (3) Å  
 c = 19.948 (5) Å  
 β = 91.461 (7)°  
 V = 1108.4 (6) Å<sup>3</sup>  
 Z = 2

D<sub>x</sub> = 1.475 Mg m<sup>-3</sup>  
 Mo Kα radiation  
 Cell parameters from 4502 reflections  
 θ = 3.1–27.5°  
 μ = 0.72 mm<sup>-1</sup>  
 T = 123 K  
 Prism, red-violet  
 0.20 × 0.05 × 0.05 mm

Data collection

Rigaku/MSM Mercury CCD diffractometer  
 ω scans  
 Absorption correction: multi-scan (Jacobson, 1998)  
 T<sub>min</sub> = 0.783, T<sub>max</sub> = 0.964  
 8886 measured reflections

2511 independent reflections  
 2006 reflections with F<sup>2</sup> > 2σ(F<sup>2</sup>)  
 R<sub>int</sub> = 0.039  
 θ<sub>max</sub> = 27.5°  
 h = -7 → 7  
 k = -12 → 12  
 l = -25 → 25

Refinement

Refinement on F  
 R = 0.039  
 wR = 0.069  
 S = 1.07  
 2506 reflections  
 151 parameters

H-atom parameters constrained  
 w = 1/[σ<sup>2</sup>(F<sub>o</sub>) + 0.00168|F<sub>o</sub>|<sup>2</sup>]  
 (Δ/σ)<sub>max</sub> < 0.001  
 Δρ<sub>max</sub> = 0.33 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.27 e Å<sup>-3</sup>

Table 1

Selected geometric parameters (Å, °).

Fe—O1	2.032 (2)	Fe—N1	2.240 (2)
Fe—O3	2.154 (2)		
O1—Fe—O3	92.01 (8)	O3—Fe—N1	93.82 (8)
O1—Fe—N1	77.30 (8)		

Table 2

Hydrogen-bonding geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
C9—H6...O1 <sup>i</sup>	0.96	2.27	3.152 (3)	153
O3—H7...O1 <sup>ii</sup>	0.96	2.51	3.192 (3)	128
O3—H7...O2 <sup>ii</sup>	0.96	1.74	2.694 (3)	172
C4—H2...O2 <sup>iii</sup>	0.96	2.50	3.359 (3)	149
C6—H3...O2 <sup>iii</sup>	0.96	2.57	3.410 (3)	146

Symmetry codes: (i) -x, 1 - y, -z; (ii) 1 - x, 1 - y, -z; (iii) ½ - x, y - ½, ½ - z.

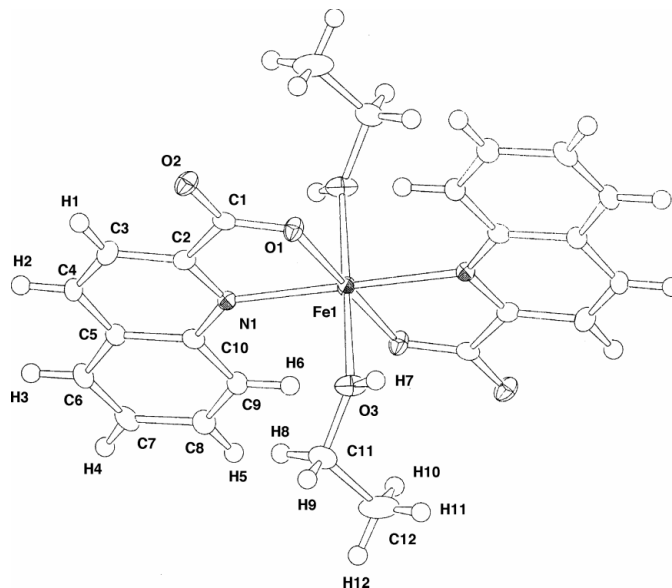


Figure 1

ORTEP-3 drawing (Farrugia, 1997) of (I), half of which defines the asymmetric unit, showing the atomic numbering scheme. Displacement ellipsoids for the non-H atoms are drawn at the 50% probability level.

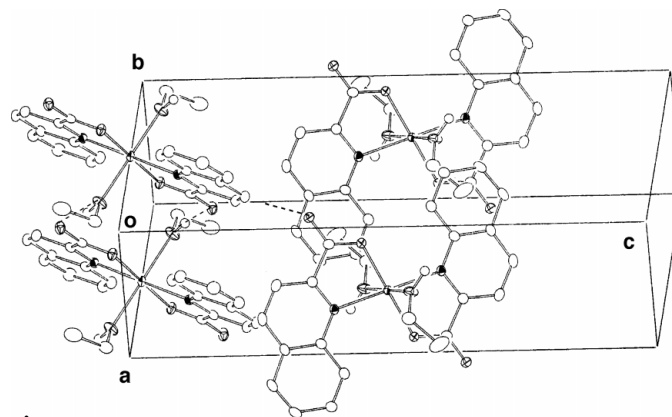


Figure 2

Packing diagram of the title complex. Dotted lines show hydrogen bonding, which forms molecular chains parallel to the a axis.

program(s) used to refine structure: TEXSAN; molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: TEXSAN.

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Hyphen	-	
Rule	⊗/	
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Capitals	<u>Cap</u>	≡ } = } — } ~ } under characters
Small capitals	<u>Sc</u>	
Italic type	<u>Ital</u>	
Bold type	<u>Bold</u>	
Lower case letters	<u>Lc</u>	} Circle characters
Roman type	<u>Rom</u>	
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Close up	∩	∩ around space to be closed up
Invert type	⊖	Circle inverted characters
Transpose	⊖	⊖ between letters or words
Faulty setting (e.g. broken type)	X	Circle defective characters
Leave as printed	<u>stet</u>	..... under material to be left
New paragraph	<u>n.p.</u>	⌈ before first word of new paragraph
No new paragraph or line	<u>run on</u>	↪ between lines

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