## 40. A Natural Pyroxene with the Space Group $C_{2h}^4$ -P2/n

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(Comm. by Tei-ichi Ito, M. J. A., Feb. 12, 1970)

Single-crystal X-ray diffraction studies of clinopyroxenes were carried out by Warren and Bragg<sup>1</sup>) and Warren and Biscoe,<sup>2</sup>) who showed that clinopyroxenes belong to the same space group  $C_{2b}^{6}$ -C2/c. On the basis of the twinned space group theory, Ito<sup>3</sup> suggested that there might exist a kind of clinopyroxenes with space group  $C_{2h}^{5}$ -P2<sub>1</sub>/c. Morimoto<sup>4),5)</sup> found that the symmetry of pigeonite and clinoenstatite is actually  $P2_1/c$ . Lindemenn<sup>6),7)</sup> reported later a clinoenstatite that has  $C_{2h}^{5}$ -P2<sub>1</sub>/n symmetry. Recently spodumene was found to have  $C_{2}^{3}$ -C2 symmetry by Appleman and Stewart,<sup>8)</sup> and the eclogite clinopyroxene omphacite was considered to have  $C_2^1$ -P2 symmetry by Clark and Papike.<sup>9),10)</sup> Thus five kinds of space groups in clinopyroxenes—C2/c  $P2_1/c$ ,  $P2_1/n$ , C2 and P2—, have been reported. Theoretically, many space groups as the subgroups of  $C_{2h}^6$ -C2/c in clinopyroxenes are possible (Matsumoto<sup>11)</sup>). In the course of study of pyroxenes in our laboratory a new type of clinopyroxene having primitive lattice with n-glide planes perpendicular to [010] in diopside setting, was found.

The material is the omphacite from a hornblende-bearing eclogite of the Iratu mass, Bessi area, Japan, which is a low temperature eclogite belonging to the epidote amphibolite facies (Banno<sup>12</sup>). The petrography of the rock will be described elsewhere by Banno and Oki.<sup>13)</sup> The chemistry of the omphacite is SiO<sub>2</sub>=53.01, TiO<sub>2</sub>=0.19, Al<sub>2</sub>O<sub>3</sub>=11.25, Fe<sub>2</sub>O<sub>3</sub>=5.04, FeO=2.56, MnO=0.02, MgO=7.34, CaO =13.21, Na<sub>2</sub>O=6.90, K<sub>2</sub>O=0.04, H<sub>2</sub>O(+)=0.25, H<sub>2</sub>O(-)=0.09, P<sub>2</sub>O<sub>5</sub> =0.01, Total=99.91. The atomic ratio for 6 oxygens (anhydrous basis) is as follows: Si=1.918, Al<sup>1v</sup>=0.082, Al<sup>v1</sup>=0.398, Fe<sup>3+</sup>=0.137, Fe<sup>2+</sup>=0.077, Ti=0.005, Mn=0.001, Mg=0.392, Ca=0.516, Na=0.484, K=0.001. Thus it is a typical omphacite containning appreciable amount of acmite component.

The Weissenberg photographs were taken about the b- and c-axes, and [110]-direction of the omphacite, up to the second layer, using Cu  $K\alpha$  radiation with Ni filter. The unit cell dimensions of omphacite are as follows;

$$a = 9.59 \pm 0.01 \text{ Å}$$
  
 $b = 8.76 \pm 0.01 \text{ Å}$ 

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$$\mathrm{c}{=}~~5.24{\pm}0.02$$
 Å $eta{=}106.9~+0.2^\circ$ 

These values are in rather good agreement with those obtained by Clark and Papike.

The extinction rules of the X-ray reflections of the omphacite are definitely not those of  $C_{2h}^6$ -C2/c, nor  $C_{2h}^5$ -P2<sub>1</sub>/c, the reflections of hkl being present even when h + k is odd, those of 0k0 also being present even when k is odd. In order to clarify the existence of glide planes, the Weissenberg h0l photograph was exposed for 200 hours. The photograph and the corresponding diagram are shown in Figs. 1 and 2 respectively. Careful examinations of this photograph revealed the

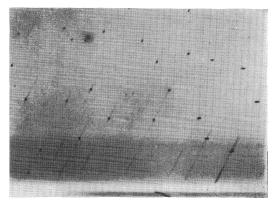


Fig. 1. The Weissenberg photograph of omphacite (rotation about b-axis, zero-layer). Reflections with h+l=2n are observed.

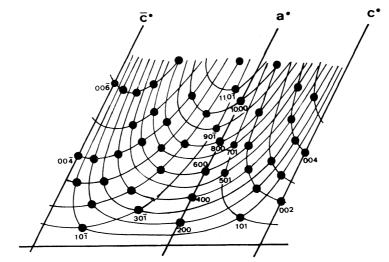


Fig. 2. The Weissenberg diagram of omphacite, corresponding to Fig. 1.

fact that the reflections of h0l are present only when h+l=2n. The same extinction rule was confirmed by counter method on another omphacite crystal.

Therefore it is concluded that the omphacite has n-glide planes, the diffraction symbol is 2/m P-/n and the possible space group is  $C_{2h}^4$ -P2/n or  $C_s^2$ -Pn. It seems more plausible that the space group is P2/n, rather than Pn. If the omphacite structure is derived from diposide type C2/c by the ordering of metallic cations, the subgroups of space group will become lower with the increasing degree of orderness, so that P2/n, that represents higher order of subgroup than Pn does, may be realized more easily than the other.

The structure analysis of this n-glide omphacite is now in progress here in cooperation with N. Morimoto and M. Tokonami, Osaka Univ. and A. Kawahara, Okayama Univ.

The writers wish to express their sincere thanks to Professor T. Ito, M. J. A., for the critical reading of the manuscript, and to Prof. Y. Matsui, Okayama Univ. for valuable discussions. Assistance by Messrs. I. Kawai and T. Nakamichi in taking the photograph is also appreciated.

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