A novel reductive amino－cyclization method and its application for the total syntheses of $( \pm)$－aurantio－clavine and（ $\pm$ ）－lophocerine

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# A NOVEL REDUCTIVE AMINO-CYCLIZATION METHOD AND ITS APPLICATION FOR THE TOTAL SYNTHESES OF ( $\pm$ )-AURANTIOCLAVINE AND ( $\pm$ )-LOPHOCERINE ${ }^{\text {1a\# }}$ 

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#### Abstract

A novel reductive amino-cyclization method for the synthesis of azacycloalkanes is developed. Its versatility is proved by the total syntheses of ( $\pm$ )-aurantioclavine (1), an ergot alkaloid, and ( $\pm$ )-lophocerine (2), a cactus alkaloid, as examples of azepane and piperidine skeletons, respectively.


In our project for developing biologically active compounds, ${ }^{2}$ we needed a novel simple method for constructing azacycloalkanes under acidic conditions. Generally speaking, however, the formation of a $\mathrm{C}-\mathrm{N}$ bond between nitrogen atom of amines and carbocation generated in situ by the action of an acid is impossible, because the basic nitrogen is protonated completely and eventually loses the nucleophilic ability.

To overcome the problem we have conceived the following idea as shown in general formula in Scheme 1. If the type A compound, having both nitro and unsaturated groups in a molecule, is subjected to an acidic reduction (in this case, $\mathrm{E}^{+}=\mathrm{H}^{+}$), the nitrogen atoms of both the resulting intermediates, hydroxylamine (B) or oxime (C), would not be protonated completely due to the attached electron-attracting oxygen atom. So, in cases where a carbocation is generated in the neighborhood of the nitrogen atom, both carbon and nitrogen atoms can connect together resulting in the formation of a cyclic hydroxylamine or a cyclic nitrone (D). Simultaneous acidic reduction can provide the desired type $\mathbf{E}$ azacycloalkanes.
On the basis of the above idea, we have succeeded in creating a novel reductive amino-cyclization method. ${ }^{3}$ We wish to report its application for the total syntheses of ( $\pm$ )-aurantioclavine ${ }^{4}$ ( $\mathbf{1}$ ), a member of ergot alkaloid and ( $\pm$ )-lophocerine (2), a cactus alkaloid, as examples of azepane and a piperidine forma-
\# Dedicated to the $75^{\text {th }}$ birthday of Prof. Dr. Ekkelhard Winterfeldt.
tions, respectively (Scheme 2). Part of these works is reported as preliminary communications. ${ }^{3,6}$ As for an example of pyrrolidine formation, we had already applied the method for the synthesis of 6,6a,7,8,9,9a-hexahydro- $2 H$-isoindolo[4,5,6-cd]indole ${ }^{3,6}$ (3), a synthetic intermediate for $\alpha$-cyclopiazonic acid (4). We had demonstrated as well that the process, $\mathbf{C} \rightarrow \mathbf{D}$ in Scheme 1, actually worked. ${ }^{7}$ Thus, the compound (5) was successfully converted to a nitrone (6, in this case, $\mathrm{E}^{+}=\mathrm{Br}^{+}$), an important intermediate for the synthesis of gibberellin $\mathrm{A}_{15}(7) .{ }^{7}$

Scheme 1. $\mathrm{S}^{1}-\mathrm{S}^{3}$ : an appropriate substituent. $\mathrm{E}^{+}$: an electrophile.


## An Efficient Total Synthesis of ( $\pm$ )-Aurantioclavine (1)

According to our synthetic procedures, ${ }^{6,8}$ we first prepared ( $E$ )-2-methyl-4-[3-(2-nitroethyl)indol-4-yl]-3-buten-2-ol (9, Scheme 2), our common synthetic intermediate for the ergot alkaloids, ${ }^{2}$ in $46 \%$ overall yield from indole-3-carbaldehyde (8) in four steps. ${ }^{8}$ Since the compound (9) contains the type $\mathbf{A}$ structure, the desired allyl cation being a benzyl cation as well, can be generated on the side chain at the 4-position of $\mathbf{9}$ by an acid treatment. The carbocation is so stable that it would survive during the time when the acidic reduction of the nitro group on the side chain at the 3 -position of 9 proceeds, and it would eventually have a chance to form a bond to the hydroxylamine nitrogen.

## Scheme 2


1: aurantioclavine

10

2: lophocerine

(Hg), МеОН-HCl


8

4

5


In fact, however, treatment of $\mathbf{9}$ with such reagent systems as $\mathrm{SnCl}_{2} / \mathrm{HCl} / \mathrm{MeOH}, \mathrm{TiCl}_{3} / \mathrm{MeOH}, \mathrm{Fe} / \mathrm{AcOH}$, and $\mathrm{Zn} / \mathrm{AcOH}$ ended in failure under various examined reaction conditions. Finally, we tried amalgamated zinc $(\mathrm{Zn}(\mathrm{Hg}))$ and found it a reagent of choice. Thus, treatment of 9 with $\mathrm{Zn}(\mathrm{Hg})$ in refluxing methanolic HCl gave ( $\pm$ )-aurantioclavine (1) in $72 \%$ yield. Further treatment of $\mathbf{1}$ with $\mathrm{Ac}_{2} \mathrm{O}$ and pyridine afforded ( $\pm$ )- $N$-acetylaurantioclavine (10) in $93 \%$ yield. Spectral data of 1 were identical with those of aurantioclavine, reported in the literature. ${ }^{4}$

Thus, we accomplished the first total synthesis of ( $\pm$ )-aurantioclavine (1) in the simplest five steps with a $33 \%$ overall yield from 8 without using any protective groups. Hegedus' group ${ }^{9}$ achieved the total syntheses of $( \pm)$ - $\mathbf{1}$ afterwards, too. ${ }^{10}$

## Total Synthesis of ( $\pm$ )-Lophocerine (2)

In order to examine whether our amino-cyclization method works for the piperidine formation, we selected ( $\pm$ )-lophocerine (2) ${ }^{5}$ as a target molecule, which is known to exist as a racemate. ${ }^{5}$

For the synthesis of intermediates having 1 -substituted isoquinoline skeletons (12a,b, Scheme 3), we chose 4-[4,5-dimethoxy- (11a) and 4-[5-benzyloxy-4-methoxy-2-(2-nitroethyl)phenyl]-2-meth-yl-3-buten-2-ol ${ }^{11}$ (11b) as suitable substrates because they have the type A structure (Scheme 1) in the molecules.

## Scheme 3

a) $\mathrm{R}=\mathrm{Me}$; b) $\mathrm{R}=\mathrm{CH}_{2} \mathrm{Ph}$


Employing the reduction with $\mathrm{NaBH}_{4}$ in MeOH , compounds (11a, ${ }^{11}$ ) were easily prepared in 74 and $81 \%$ yields, respectively, from 13a and 13b. ${ }^{11}$ They were prepared according to our procedures ${ }^{11}$ from 6-bromoveratraldehyde (14a) and 6-bromo-4-benzyloxy-3-methoxybenzaldehyde (14b) in two steps through 15a and 15b.

The reductive amino-cyclization method, using zinc instead of $\mathrm{Zn}(\mathrm{Hg})$, was successfully applied for 11a and 11b. Thus, treatment of 11a and 11b with zinc in refluxing methanolic HCl provided 6,7-dimethoxy-
(12a) and 7-benzyloxy1-6-methoxy-1-(2-methyl-1-propen-1-yl)-1,2,3,4-tetrahydroisoquinoline (12b) in 67 and $77 \%$ yields, respectively.

Further treatment of $\mathbf{1 2 b}$ with $\mathrm{ClCO}_{2} \mathrm{Me}$ and $\mathrm{Et}_{3} \mathrm{~N}$ afforded 7-benzyloxy-6-methoxy-2-methoxy-carbonyl-1-(2-methyl-1-propen-1-yl)-1,2,3,4-tetrahydroisoquinoline (16) in $99 \%$ yield. Subsequent reduction of 16 with $\mathrm{LiAlH}_{4}$ in dry THF gave 7-benzyloxy-1-(2-methyl-1-propen-1-yl)-6-meth-oxy-2-methyl-1,2,3,4-tetrahydroisoquinoline (17) in $86 \%$ yield together with an $11 \%$ yield of the debenzylated compound 18. Catalytic hydrogenation of the double bond and debenzylation of 17 over $10 \% \mathrm{Pd} / \mathrm{C}$ in MeOH produced ( $\pm$ )-lophocerine (2) in 39\% yield. Our synthesized 2 and its picrate were identical with the samples prepared according to the procedure described in the literature. ${ }^{5}$
In conclusion, we have established a novel reductive amino-cyclization method for the synthesis of azacycloalkanes based on our idea shown in Scheme 1. Its versatility was proved by the total syntheses of ( $\pm$ )-aurantioclavine (1) and ( $\pm$ )-lophocerine (2) as examples of the azepane and piperidine skeletons, respectively.

## EXPERIMENTAL

Melting points were determined on a Yanagimoto micro melting point apparatus and are uncorrected. IR spectra were determined with a Shimadzu IR-420 spectrophotometer, and ${ }^{1} \mathrm{H}$-NMR spectra with a JEOL JNM-PMX 60 or a JEOL FX-100 spectrometer, with tetramethylsilane as an internal standard. MS were recorded on a Hitachi M-80 spectrometer. Preparative thin-layer chromatography (p-TLC) was performed on Merck Kiesel-gel $\mathrm{GF}_{245}$ (Type 60) ( $\mathrm{SiO}_{2}$ ). Column chromatography was performed on silica gel ( $\mathrm{SiO}_{2}, 100-200$ mesh, from Kanto Chemical Co., Inc.) throughout the present study.
( $\pm$ )-Aurantioclavine (1) from (E)-2-Methyl-4-[3-(2-nitroethyl)indol-4-yl]-3-buten-2-ol (9) — A solution of 9 ( $113.2 \mathrm{mg}, 0.41 \mathrm{mmol}$ ) in $\mathrm{MeOH}(24 \mathrm{~mL})$ and $3 \% \mathrm{HCl}(8 \mathrm{~mL})$ was added to $\mathrm{Zn}(\mathrm{Hg})$, prepared from Zn powder ( $1.386 \mathrm{~g}, 21.2 \mathrm{mmol}$ ) and $\mathrm{HgCl}_{2}$ ( $229.0 \mathrm{mg}, 0.84 \mathrm{mmol}$ ) in $3 \% \mathrm{HCl}(8 \mathrm{~mL})$, and the mixture was refluxed for 3.5 h with stirring. Unreacted $\mathrm{Zn}(\mathrm{Hg})$ was filtered off and the filtrate was evaporated under reduced pressure. The residue was made basic by adding $8 \% \mathrm{NaOH}$ and the whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(95: 5, \mathrm{v} / \mathrm{v})$. The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave an oil, which was subjected to p-TLC on $\mathrm{SiO}_{2}$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}(46: 5: 0.5, \mathrm{v} / \mathrm{v})$ as a developing solvent. Extraction of the band having an $R f$ value of $0.29-0.58$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}(46: 5: 0.5$, v/v) gave $\mathbf{1}$ ( $66.9 \mathrm{mg}, 72 \%$ ). 1: $\mathrm{mp} 194-196^{\circ} \mathrm{C}$ (colorless leaves, recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-hexane). IR ( KBr ): 3290, $2915 \mathrm{~cm}^{-1}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 1.71\left(1 \mathrm{H}, \mathrm{br}\right.$ s, disappeared on addition of $\left.\mathrm{D}_{2} \mathrm{O}\right), 1.84(6 \mathrm{H}, \mathrm{s}), 2.83-3.24(3 \mathrm{H}, \mathrm{m})$, $3.33-3.71(1 \mathrm{H}, \mathrm{m}), 4.86(1 \mathrm{H}, \mathrm{d}, J=9.0 \mathrm{~Hz}), 5.44(1 \mathrm{H}$, br d, $J=9.0 \mathrm{~Hz}), 6.80(1 \mathrm{H}$, br d, $J=6.8 \mathrm{~Hz})$,
6.89—7.27(2H, m), $6.94(1 \mathrm{H}, \mathrm{br} \mathrm{s}), 8.17\left(1 \mathrm{H}, \mathrm{br}\right.$ s, disappeared on addition of $\left.\mathrm{D}_{2} \mathrm{O}\right) . \mathrm{MS} \mathrm{m} / \mathrm{z}$ : $226\left(\mathrm{M}^{+}\right)$. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}_{2}$ : C, 79.60; H, 8.02; N, 12.38. Found: C, 79.52; H, 8.10; N, 12.53.
$\mathbf{(} \mathbf{)}$ - N -Acetylaurantioclavine (10) from $\mathbf{1}$ - $\mathrm{Ac}_{2} \mathrm{O}(1 \mathrm{~mL})$ was added to a solution of $\mathbf{1}(53.1 \mathrm{mg}, 0.24$ mmol ) in pyridine ( 2 mL ) and the mixture was stirred at rt for 13 h . After evaporation of the solvent, saturated aqueous $\mathrm{NaHCO}_{3}$ was added and the whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(95: 5, \mathrm{v} / \mathrm{v})$. The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave a solid, which was recrystallized from MeOH to give $\mathbf{1 0}$ ( $58.5 \mathrm{mg}, 93 \%$ ) as colorless needles. 10: mp $235-236^{\circ} \mathrm{C}$. IR (KBr): 3200, 1664, $1593 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(10 \% \mathrm{CD}_{3} \mathrm{OD}\right.$ in $\left.\mathrm{CDCl}_{3}\right) \delta: 1.70(3 \mathrm{H}, \mathrm{s}), 1.80$ $(3 \mathrm{H}, \mathrm{s}), 2.18(3 \mathrm{H}, \mathrm{s}), 2.64-4.54(4 \mathrm{H}, \mathrm{m}), 5.72(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=7.5 \mathrm{~Hz}), 5.80(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=7.5 \mathrm{~Hz})$, 6.57-7.28 (4H, m). MS m/z: $268\left(\mathrm{M}^{+}\right)$. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: \mathrm{C}, 76.08 ; \mathrm{H}, 7.51 ; \mathrm{N}, 10.44$. Found: C, 75.90; H, 7.59; N, 10.22.
(E)-4-[4,5-Dimethoxy-2-(2-nitroethyl)phenyl]-2-methyl-3-buten-2-ol (11a) from (E,E)-4-[4,5-Dimethoxy-2-(2-nitrovinyl)phenyl]-2-methyl-3-buten-2-ol (13a) - $\mathrm{NaBH}_{4}$ (1.045g, 27.5 mmol ) was added to a solution of $\mathbf{1 3 a}(2.009 \mathrm{~g}, 6.86 \mathrm{mmol})$ in $\mathrm{MeOH}(120 \mathrm{~mL})$ and the mixture was stirred at rt for 30 min . Brine was added and the mixture was adjusted to pH 8 by adding $6 \% \mathrm{HCl}$. The whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (95:5, v/v). The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave an oil, which was column-chromatographed on $\mathrm{SiO}_{2}$ with $\mathrm{Et}_{2} \mathrm{O}$-hexane ( $3: 1, \mathrm{v} / \mathrm{v}$ ) to give 11a ( $1.245 \mathrm{~g}, 62 \%$ ). 11a: mp $99-100^{\circ} \mathrm{C}$ (colorless prisms, recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-hexane). IR (KBr): 3460, 1602, 1538, 1503, 1343, $1261 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right) \delta: 1.42(6 \mathrm{H}, \mathrm{s}), 1.83\left(1 \mathrm{H}, \mathrm{br}\right.$ s, disappeared on addition of $\left.\mathrm{D}_{2} \mathrm{O}\right), 3.27(2 \mathrm{H}, \mathrm{t}, J=7.2 \mathrm{~Hz}), 3.78$ $(3 \mathrm{H}, \mathrm{s}), 3.81(3 \mathrm{H}, \mathrm{s}), 4.43(2 \mathrm{H}, \mathrm{t}, J=7.2 \mathrm{~Hz}), 6.03(1 \mathrm{H}, \mathrm{d}, J=15.6 \mathrm{~Hz}), 6.52(1 \mathrm{H}, \mathrm{s}), 6.64(1 \mathrm{H}, \mathrm{d}, J=15.6$ $\mathrm{Hz}), 6.82(1 \mathrm{H}, \mathrm{s})$. MS m/z: $295\left(\mathrm{M}^{+}\right)$. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{5} \cdot 1 / 10 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 60.63$; H, 7.19; N, 4.71. Found: C, 60.61; H, 7.27; N, 4.79.

6,7-Dimethoxy-1-(2-methyl-1-propen-1-yl)-1,2,3,4-tetrahydroisoquinoline (12a) from 11a - A solution of $\mathbf{1 1 a}(50.9 \mathrm{mg}, 0.17 \mathrm{mmol})$ in $\mathrm{MeOH}(6 \mathrm{~mL})$ was added to a mixture of Zn powder in $6 \% \mathrm{HCl}$ ( 2 mL ), and the mixture was refluxed for 3 h with stirring. Unreacted Zn was filtered off and the filtrate was evaporated under reduced pressure. The residue was made basic by adding $8 \% \mathrm{NaOH}$ and the whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(95: 5, \mathrm{v} / \mathrm{v})$. The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave an oil, which was column-chromatographed on $\mathrm{SiO}_{2}$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \% \mathrm{aq} . \mathrm{NH}_{3}$ (46:5:0.5, v/v) to give 12a ( $28.7 \mathrm{mg}, 67 \%$ ). 12a: mp $57-58^{\circ} \mathrm{C}$ (colorless needles, recrystallized from petroleum ether). IR (KBr): 3400, 2910, 2760, 1605, 1509, 1259 $\mathrm{cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 1.80(3 \mathrm{H}, \mathrm{d}, J=1.2 \mathrm{~Hz}), 1.84(3 \mathrm{H}, \mathrm{d}, J=1.2 \mathrm{~Hz}), 2.04(1 \mathrm{H}, \mathrm{br}$ s, disappeared on addition of $\left.\mathrm{D}_{2} \mathrm{O}\right), 2.44-3.40(4 \mathrm{H}, \mathrm{m}), 3.80(3 \mathrm{H}, \mathrm{s}), 3.84(3 \mathrm{H}, \mathrm{s}), 4.65(1 \mathrm{H}, \mathrm{d}, J=9.3 \mathrm{~Hz}), 5.28(1 \mathrm{H}, \mathrm{br} \mathrm{d}$,
$J=9.3 \mathrm{~Hz}), 6.51(1 \mathrm{H}, \mathrm{s}), 6.57(1 \mathrm{H}, \mathrm{s})$. MS m/z: $247\left(\mathrm{M}^{+}\right)$. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{2}$ : C, 72.84; H, 8.56; N, 5.66. Found: C, 72.62; H, 8.66; N, 5.59.
7-Benzyloxy-6-methoxy-1-(2-methyl-1-propen-1-yl)-1,2,3,4-tetrahydroisoquinoline (12b) from 11b

- A solution of $\mathbf{1 1 b}$ ( $36.2 \mathrm{mg}, 0.09 \mathrm{mmol}$ ) in $\mathrm{MeOH}(6 \mathrm{~mL})$ was added to a mixture of Zn powder ( $332.1 \mathrm{mg}, 5.08 \mathrm{mmol}$ ) in $6 \% \mathrm{HCl}(2 \mathrm{~mL}$ ) and the mixture was refluxed for 4 h with stirring. Unreacted Zn was filtered off and the filtrate was evaporated under reduced pressure. The residue was made basic by adding $20 \% \mathrm{NaOH}$ and the whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(95: 5, \mathrm{v} / \mathrm{v})$. The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave an oil, which was subjected to p-TLC on $\mathrm{SiO}_{2}$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}$ (46:5:0.5, v/v) as a developing solvent. Extraction of the band having an $R f$ value of $0.49-0.72$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}$ (46:5:0.5, v/v) gave 12b ( $24.2 \mathrm{mg}, 77 \%$ ). 12b: mp $114-116^{\circ} \mathrm{C}$ (colorless prisms, recrystallized from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ). IR (KBr): 3230, 2900, 1604, 1502, 1359, 1258, 1245, $1215 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta:$ $1.71\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J}=2 \mathrm{~Hz}\right.$ ), $1.89\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}\right.$, disappeared on addition of $\left.\mathrm{D}_{2} \mathrm{O}\right), 2.46-2.89(2 \mathrm{H}, \mathrm{m}), 2.89-3.33$ $(2 \mathrm{H}, \mathrm{m}), 3.77(3 \mathrm{H}, \mathrm{s}), 4.46(1 \mathrm{H}, \mathrm{d}, J=9 \mathrm{~Hz}), 4.97(2 \mathrm{H}, \mathrm{s}), 5.07(1 \mathrm{H}, \mathrm{br} \mathrm{d}, \mathrm{J}=9 \mathrm{~Hz}), 6.36(1 \mathrm{H}, \mathrm{s}), 6.44(1 \mathrm{H}$, s), $7.02-7.45(5 \mathrm{H}, \mathrm{m})$. MS m/z: $323\left(\mathrm{M}^{+}\right)$. Anal. Calcd for $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{NO}_{2}$ : C, 77.98; H, 7.79; N, 4.33. Found: C, 77.69; H, 7.83; N, 4.20.


## 7-Benzyloxy-6-methoxy-2-methoxycarbonyl-1-(2-methyl-1-propen-1-yl)-1,2,3,4-tetrahydroisoquino

 line (16) from 12b - A solution of $\mathrm{ClCO}_{2} \mathrm{Me}(45.4 \mathrm{mg}, 0.48 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ and $\mathrm{Et}_{3} \mathrm{~N}(0.2$ $\mathrm{mL})$ were added to a solution of $\mathbf{1 2 b}(82.4 \mathrm{mg}, 0.26 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ and the mixture was stirred at rt for 3 h . Saturated aqueous $\mathrm{NaHCO}_{3}$ was added and the whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (95:5, v/v). The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave an oil, which was subjected to p-TLC on $\mathrm{SiO}_{2}$ with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(99: 1, \mathrm{v} / \mathrm{v})$ as a developing solvent. Extraction of the band having an $R f$ value of $0.34-0.68$ with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (95:5, $\mathrm{v} / \mathrm{v}$ ) gave 16 ( $96.6 \mathrm{mg}, 99 \%$ ). 16: Colorless oil. IR (film): 1697, $1608 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 1.67$ (3H, d, J=1.2 Hz), 1.83 (3H, d, J=1.2 Hz), 2.21—4.37 (4H, m), 3.61 (3H, s), 3.73 (3H, s), 4.91 ( $2 \mathrm{H}, \mathrm{s}$ ), $5.08(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=9.5 \mathrm{~Hz}), 5.49(1 \mathrm{H}, \mathrm{d}, J=9.5 \mathrm{~Hz}), 6.31(1 \mathrm{H}, \mathrm{s}), 6.39(1 \mathrm{H}, \mathrm{s}), 7.01-7.42(5 \mathrm{H}, \mathrm{m})$. High-resolution MS m/z: Calcd for $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{4}$ : 381.1938. Found: 381.1956.7-Benzyloxy- (17) and 7-Hydroxy-6-Methoxy-2-methyl-1-(2-methyl-1-propen-1-yl)-1,2,3,4tetrahydroisoquinoline (18) from $16-\mathrm{LiAlH}_{4}(109.2 \mathrm{mg}, 0.29 \mathrm{mmol})$ was added to a solution of $\mathbf{1 6}$ ( $96.6 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) in anhydrous THF ( 6 mL ) and the mixture was refluxed for 24 h . After the addition of MeOH and saturated aqueous Rochelle salt under ice cooling, the whole was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(95: 5, \mathrm{v} / \mathrm{v})$. The organic layer was washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure to leave an oil, which was subjected to p-TLC on $\mathrm{SiO}_{2}$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}(46: 5: 0.5, \mathrm{v} / \mathrm{v})$ as a developing solvent. Extraction of the band having an $R f$ value of $0.37-0.69$
with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}$ (46:5:0.5, v/v) gave 17 ( $73.8 \mathrm{mg}, 86 \%$ ). 17: Colorless oil. IR (film): 2930, 2780, 1610, 1507, $1252 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 1.67(3 \mathrm{H}, \mathrm{d}, J=1.2 \mathrm{~Hz}), 1.85(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.5 \mathrm{~Hz})$, $2.15-3.28(4 \mathrm{H}, \mathrm{m}), 2.28(3 \mathrm{H}, \mathrm{s}), 3.72(1 \mathrm{H}, \mathrm{d}, J=9.5 \mathrm{~Hz}), 3.77(3 \mathrm{H}, \mathrm{s}), 4.93(1 \mathrm{H}, \mathrm{d}, J=9.5 \mathrm{~Hz}), 4.98(2 \mathrm{H}$, s), $6.35(1 \mathrm{H}, \mathrm{s}), 6.46(1 \mathrm{H}, \mathrm{s}), 7.02-7.38(5 \mathrm{H}, \mathrm{m})$. High-resolution MS m/z: Calcd for $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{NO}_{2}$ : 337.2040. Found: 337.2038. Extraction of the band having an $R f$ value of $0.20-0.29$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}\left(46: 5: 0.5\right.$, v/v) gave 18 ( $6.7 \mathrm{mg}, 11 \%$ ). 18: mp $159 — 160^{\circ} \mathrm{C}$ (colorless prisms, recrystallized from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ). IR (KBr): 2900, 1589, 1501, $1268 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ : 1.77 ( $6 \mathrm{H}, \mathrm{d}, J=1.2 \mathrm{~Hz}$ ), 2.17-3.17 (4H, m), 2.32 ( $3 \mathrm{H}, \mathrm{s}$ ), $3.74(3 \mathrm{H}, \mathrm{s}), 3.83$ ( $1 \mathrm{H}, \mathrm{d}, J=9.6 \mathrm{~Hz}$ ), 4.77 ( 1 H , br s, disappeared on addition of $\mathrm{D}_{2} \mathrm{O}$ ), $5.05(1 \mathrm{H}, \mathrm{br} \mathrm{d}, \mathrm{J}=9.6 \mathrm{~Hz})$, $6.42(2 \mathrm{H}, \mathrm{s})$. MS m/z: $247\left(\mathrm{M}^{+}\right)$. High-resolution MS m/z: Calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{2}$ : 247.1573. Found: 247.1569.
( $\pm$ )-Lophocerine (2) from 17 - A solution of 17 ( $30.9 \mathrm{mg}, 0.09 \mathrm{mmol}$ ) in $\mathrm{MeOH}(7 \mathrm{~mL})$ was hydrogenated in the presence of $10 \% \mathrm{Pd} / \mathrm{C}(33.7 \mathrm{mg})$ at rt and 1 atm for 7 h . Catalyst was filtered off and the filtrate was evaporated under reduced pressure to leave an oil, which was purified by p-TLC with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \% \mathrm{aq} . \mathrm{NH}_{3}(46: 5: 0.5, \mathrm{v} / \mathrm{v})$ as a developing solvent. Extraction of the band having an $R f$ value of $0.37-0.50$ with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-28 \%$ aq. $\mathrm{NH}_{3}$ (46:5:0.5, v/v) gave 2 ( $8.8 \mathrm{mg}, 39 \%$ ). 2: Colorless oil. IR (film): 2940, 1592, 1505, 1461, 1360, 1266, 1210, 1130, 1102, 1021, 869, $775 \mathrm{~cm}^{-1}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.90(3 \mathrm{H}, \mathrm{d}, J=6 \mathrm{~Hz}), 0.95(3 \mathrm{H}, \mathrm{d}, J=6 \mathrm{~Hz}), 1.22-2.02(3 \mathrm{H}, \mathrm{m}), 2.17-3.57(5 \mathrm{H}$, $\mathrm{m}), 2.40(3 \mathrm{H}, \mathrm{s}), 3.80(3 \mathrm{H}, \mathrm{s}), 4.33\left(1 \mathrm{H}\right.$, br s, disappeared on addition of $\left.\mathrm{D}_{2} \mathrm{O}\right), 6.43(1 \mathrm{H}, \mathrm{s}), 6.50(1 \mathrm{H}, \mathrm{s})$. MS m/z: $249\left(\mathrm{M}^{+}\right) .2 \cdot p i c r a t e: ~ m p 194-196^{\circ} \mathrm{C}$ (lit., ${ }^{5} \mathrm{mp} 191.5-193^{\circ} \mathrm{C}$, yellow prisms, recrystallized from benzene). IR (KBr): 3370, 2950, 1609, 1507, 1309, $1269 \mathrm{~cm}^{-1}$.

## REFERENCES AND NOTES

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