Flame-retardant thermoplastics derived from plant cell wall polymers by single ionic liquid substitution

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Electronic Supplementary Information

Fire-retardant thermoplastics derived from plant cell wall

by single ionic liquid substitution

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Supporting Figures

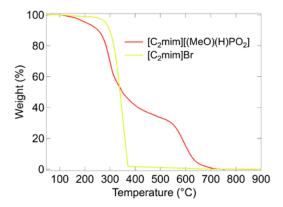


Fig. S1 TGA curves of $[C_2mim][(MeO)(H)PO_2]$ which were used in this study under air condition.

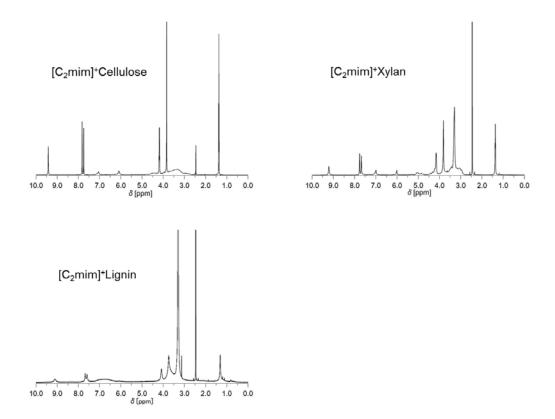


Fig. S2 ¹H NMR spectra of $[C_2 mim]^+$ cellulose, $[C_2 mim]^+$ xylan and $[C_2 mim]^+$ lignin in DMSO*d*₆.

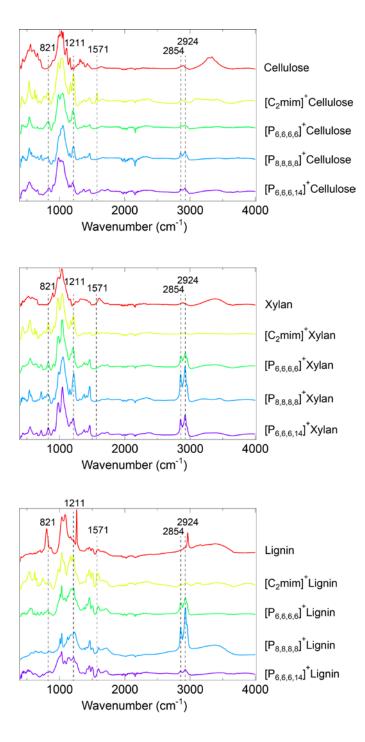


Fig. S3 FT-IR spectra of the underivatised and derivitised cellulose, xylan, and lignin.

The peaks derived from the P-O-C bond at 821 cm⁻¹, P=O bond at 1211 cm⁻¹, and C=N bond at 1571 cm⁻¹ were detected in $[C_2mim]^+$ cellulose, $[C_2mim]^+$ xylan and $[C_2mim]^+$ lignin. The peaks derived from the C-H bond of phosphonium cation at 2854 and 2924 cm⁻¹ were also detected in the spectrum of phosohonium-type derivatives. In the case of lignin derivatives, the signals are not strong due to the low derivatisation ratio.

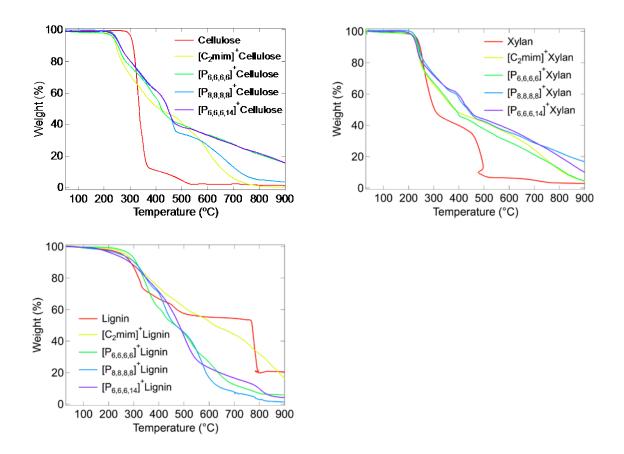


Fig. S4 TGA curves of the underivatised and derivitised cellulose, xylan, and lignin measured under air condition.

TG signals sometimes behaved abnormally (for example in the case of underivatised xylan) and the abnormal behaviour is attributed to the combustion of the samples because the samples were measured under air condition.

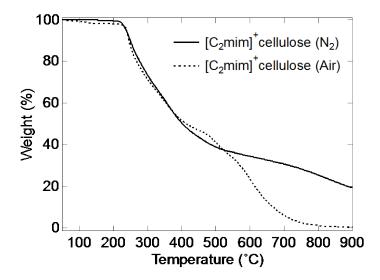


Fig. S5 TGA curves of $[C_2mim]^+$ cellulose measured under air and nitrogen gas conditions.

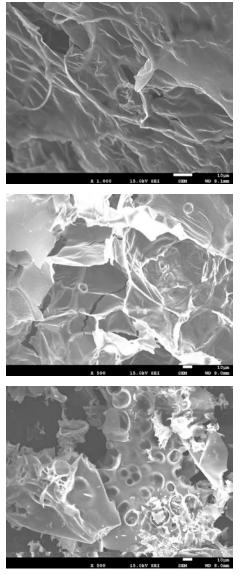


Fig. S6 SEM images of the char layers of $[C_2 \text{mim}]^+$ cellulose (top), $[C_2 \text{mim}]^+$ xylan (middle), $[C_2 \text{mim}]^+$ lignin (bottom).

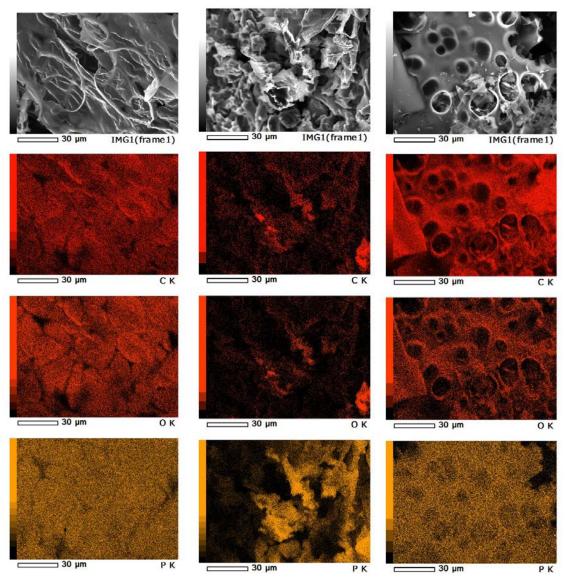


Fig. S7 SEM-EDX images of the char layers of $[C_2 mim]^+$ cellulose (top), $[C_2 mim]^+$ xylan (middle), $[C_2 mim]^+$ lignin (bottom).

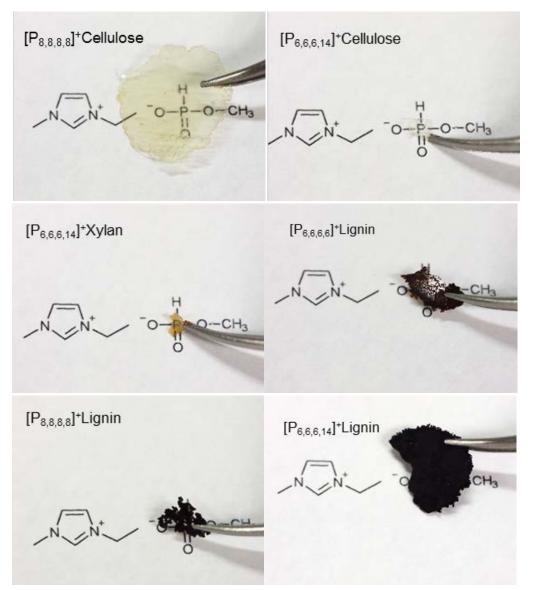


Fig. S8 Thin films of $[P_{8,8,8,8}]^+$ cellulose, $[P_{6,6,6,14}]^+$ cellulose, $[P_{6,6,6,14}]^+$ xylan, $[P_{6,6,6,6}]^+$ lignin, $[P_{8,8,8,8}]^+$ lignin, and $[P_{6,6,6,14}]^+$ lignin after hot pressing.



Fig. S9 Char formation of the thin films of $[P_{8,8,8,8}]^+$ cellulose, $[P_{6,6,6,14}]^+$ cellulose, $[P_{6,6,6,14}]^+$ xylan, $[P_{6,6,6,6}]^+$ lignin, $[P_{8,8,8,8}]^+$ lignin, and $[P_{6,6,6,14}]^+$ lignin after burning and extinguish the fire.



Fig. S10 A SEM image of the char layer of $[P_{8,8,8,8}]^+$ cellulose.

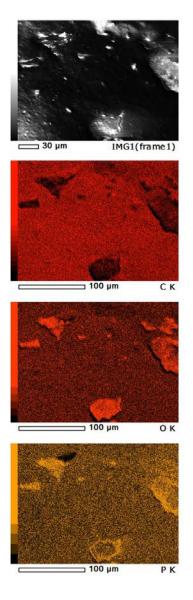


Fig. S11 A SEM-EDX image of the char layer of $[P_{8,8,8,8}]^+$ cellulose.

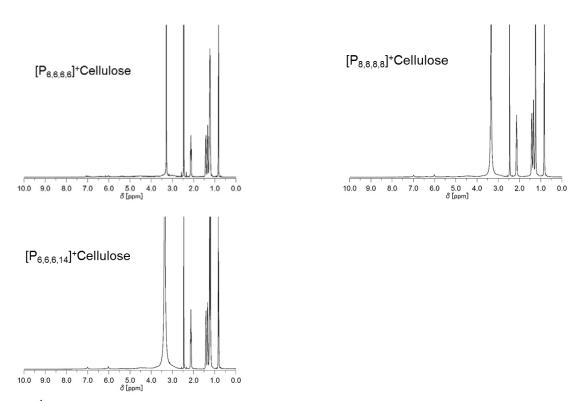


Fig. S12 ¹H NMR spectra of $[P_{6,6,6,6}]^+$, $[P_{8,8,8,8}]^+$, and $[P_{6,6,6,14}]^+$ cellulose in DMSO-*d*₆. The signal at 3.4 ppm is water.

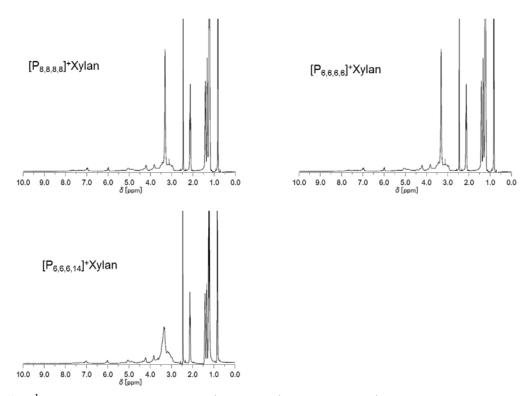


Fig. S13 ¹H NMR spectra of $[P_{6,6,6,6}]^+$, $[P_{8,8,8,8}]^+$, and $[P_{6,6,6,14}]^+$ xylan in DMSO-*d*₆. The signal at 3.4 ppm is water.

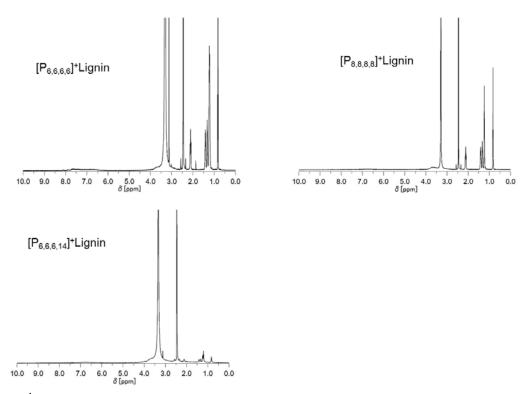


Fig. S14 ¹H NMR spectra of $[P_{6,6,6,6}]^+$, $[P_{8,8,8,8}]^+$, and $[P_{6,6,6,14}]^+$ lignin in DMSO-*d*₆. The signal at 3.4 ppm is water.