Preventive Measures and Mechanisms of Concrete Deterioration due to ASR under Marine Environments

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1991 Fiscal Year Final Research Report Summary

Preventive Measures and Mechanisms of Concrete Deterioration due to ASR under Marine Environments

Research Project

Project/Area Number
02650333
Research Category
Grant-in-Aid for General Scientific Research (C)
Allocation Type
Single-year Grants
Research Field
コンクリート工学・土木材料・施工
Research Institution
Kanazawa University
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Project Period (FY)
1990 – 1991
Keywords

Ettringite / Expansion / Exposure Tests / Opal / Alkali-Silica Reaction / Pore Solution

Research Abstract

The results obtained in this fiscal year are divided into the two categories, i. e. the one on the reaction products and expansion behavior in opalbearing mortars immersed in NaCl solutions and an artificial sea water, exposed to a marine environment. The major results obtained in the former experiments are summarized as follows ;

(1) Great amounts of Ca (OH)2 in all mortars with the opal were dissipated during the immersion in 1 N NaCl solution.

(2) Great amounts of the chloride-bearing ettringite was found in the portions as well as in voids in opal-bearing mortars immersed in 1 N NaCl

solution for longer than one vear. It is likely that the greater the expansion of mortars, the greater the amount of ettringite found in the mortars. (3) The concentration of NaCl solution in which mortar specimens were immersed, greatly influenced the expansion of mortars with relatively low

alkali/opal ratio. However, in the range of high alkali/opal ratio, the concentration of the NaCl solution little influenced the expansion of the mortars.

The following new findings were obtained in the latter experiments which have been successively carried out for these several years.

(1) Rapid expansions occurred on the exposed surfaces in reactive aggregate-bearing concrete specimens with the unit cement content of

450kg/cm3. This fact suggests that spray of sea water is influencing the expansion of the concrete due to the alkali-silica reaction.

(2) There is found a trend that the less the alkali content in the concretes, the greater the expansion of the concretes exposed to the marine environment.

Research Products (8 results)

			All	Other
	All	Publicatio	ons (8 i	esults)
[Publications] 川村 満紀,鳥居 和之,五十嵐 正巳: "海洋環境下におけるアルカリ・シリカ反応によるコンクリ-トの膨張とひびわれ" コン カリ骨材反応判定試方法に関するシンポジウム論文集. 1-6 (1991)	ンクリ-	・ト供試体に	よるアノ	ل 🗸
[Publications] 川村 満紀,杉山 彰徳,竹内 勝信: "NaCl溶液中におけるオパ-ル含有モルタルの膨張挙動と細孔溶液の組成" コンクリ-トエ (1992)	[学年次	R論 文報告集	€. 14.	~
[Publications] M.Kawamura M.Igarashi K.Takeuchi: "Relationships between Alkali/Opal Ratio,Expansion and Pore Solution Co Mortars Immersed in NaCl Solution" Proc.of 9th International Conference on alkali-Aggregate Reaction in Concrete. (1992)	omposi	ition in Op	al-Bear	ing 🗸
[Publications] M.Saito M.Kawamura: "Effects of Sodium Chloride on the Hydration Products in the Interfacial Zone between Reactive Aggregate" Proc.of International Conference on Interfacs in Cementitious Composites. (1992)	Cemer	nt Paste ar	าd Alkal	i- 🗸
[Publications] M. Kawamura, K. Torii and M. Igarashi: "Expansion and Cracks Caused by Alkali-Silica Reaction in Concrete ur Environment" Proc. of JCI Symposium on Test Method for Determination of Alkali-Aggregate Reactivity in Concrete by Using (1991)	nder a J Conci	Marine rete Specir	nen. 1-	6 🗸
[Publications] M. Kawamura, A. Sugiyama and K. Takeuchi: "Expansive Behavior and Pore Solution Composition of Opal-Bea Solution" Proc. of the JCI. 14. (1992)	aring M	lortars in N	NaCl	~
[Publications] M. Kawamura, M. Igarashi and K. Takeuchi: "Relationships between Alkali/Opal Ratio, Expansion and Pore Sol Opal-Bearing Mortars Immersed in NaCl Solution" Proc. of 9th International Conference on alkali-Aggregate Reaction in Con	lution (ncrete.	Compositic (1992)	on in	~
[Publications] M. Saito and M. Kawamura: "Effect of Sodium Chloride on the Hydration Products in the Interfacial Zone betw Alkali-Reactive Aggregate" Proc. of International Conference on Interfaces in Cementitious Composites. (1992)	veen C	ement Pas	te and	~

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