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メタデータ 言語: eng 出版者: 公開日: 2022-12-01 キーワード (Ja): キーワード (En): 作成者: メールアドレス: 所属: URL https://doi.org/10.24517/00068385

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Synthesis of 3,6-Dihydro-2*H*-1,2-oxazines via Dimethylsulfoxonium Methylide Addition to α,β-Unsaturated Nitrones

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ABSTRACT: Unique and efficient formation of 3,6-dihydro-2*H*-1,2-oxazines starting from α , β -unsaturated nitrones has been achieved. The nucleophilic addition of dimethylsulfoxonium methylide to the C=N bond of an α , β -unsaturated nitrone to form an aziridine *N*-oxide followed by Meisenheimer rearrangement affords the 3,6-dihydro-2*H*-1,2-oxazine up to 70% yield. Methylene was confirmed to be incorporated at the C₃-position of the ring. A wide range of β -aryl-substituted α , β -unsaturated nitrones was applicable to this reaction.

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INTRODUCTION

Sulfur ylides are one of the most important and widely applied reagents since the pioneering work of Corey and Chaykovsky in the 1960s (Corey-Chaykovsky reaction). These impressive successes are attributed to the inherent properties of sulfur ylides that enable them to function as nucleophilic 1,1-dipolar species. They have long been used as one-carbon synthons in transformations with different electron-deficient unsaturated functional groups, providing a variety of significant carbocyclic and heterocyclic compounds.² One of the most widespread uses of sulfur ylides is the reaction with aldehydes and ketones, which yields betaine intermediates that collapse to give epoxides. In the case of α,β -unsaturated ketones, dimethylsulfonium methylide was reported to prefer 1,2-addition while the reaction of dimethylsulfoxonium methylide tended to give the cyclopropane derivatives via 1,4-addition. 1,2,3 In the case of imines, sulfonium and sulfoxonium vlides reacted smoothly to form aziridines. 1,2 The reaction of sulfonium vlides with α , β -unsaturated imine substrates preferred 1,2-addition onto imine-carbons to produce the corresponding aziridines. However, 1,4-addition was dominant depending on the protecting group on the imine nitrogen and characteristics of the ylide.⁵ Among the imine derivatives, the reaction of sulfur ylides to nitrones has scarcely been reported. The nucleophilic addition of dimethylsulfoxonium ylide to o-hydroxy-benzaldehyde-based nitrones followed by cyclization gave dihydrobenzofuran derivatives. As part of our investigation into the reactions of 1,3-dipoles and sulfur vlides. we planned to examine the reaction of sulfur ylides to α,β -unsaturated nitrones. By the 1,3-addition to an α,β -unsaturated nitrone, aziridine I or azetidine II formation might occur by nucleophilic substitution of dimethylsulfide or dimethylsulfoxide in the betaine intermediate (pathways (i) and (ii), respectively). In contrast, cyclopropane III formation or 5,6-dihydro-2*H*-1,2-oxazine IV formation might be possible by the addition to the β -carbon of the α,β -unsaturated nitrone (1,5addition) via ring closure (pathways (iii) and (iv), respectively). The present report describes the unique formation of 3,6-dihydro-2H-1,2-oxazines by a sequential addition and rearrangement starting from α,β -unsaturated nitrones with dimethylsulfoxonium methylide via aziridine N-oxide intermediate I (Scheme 1).

Scheme 1. Possible Pathways for Sulfur Ylide-Mediated Transformations from α,β Unsaturated Nitrone

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

RESULTS AND DISCUSSION

The initial reaction of α , β -unsaturated nitrone **1a** with dimethylsulfonium methylide derived *in situ* from trimethylsulfonium iodide with NaH in THF was examined at 60 °C (oil bath temperature) for 5 h. By ¹H NMR analysis of the crude products, unreacted **1a** was recovered in 59% yield. Although several products were indicated by TLC analysis during the reaction, clear and sharp singlet signals, which could be assigned to the *N*-CH₃ group, were scarcely observed in the area from 2 to 4 ppm except *N*-methyl protons of **1a** in the ¹H NMR spectrum (Table 1, Entry 1). When dimethylsulfoxonium methylide generated *in situ* from trimethylsulfoxonium iodide and NaH was employed, most of the nitrone **1a** was consumed within 6 h. After aqueous workup, two rather sharp proton resonances were observed around 3 ppm in the ¹H NMR spectrum of the crude products. Purification by silica gel column chromatography furnished unexpected 3,6-dihydro-2*H*-1,2-oxazine **2a** in 33% yield (Entry 2). The structure of the product was confirmed by its ¹H and ¹³C NMR spectra, which were identical to the reported data. ⁸ The sharp signals at 2.71 and 3.30 ppm could be assigned to methyl protons on nitrogen and C₃ methylene protons, respectively. Using NaOH as a base instead of NaH, **2a** was obtained in only 4% yield (Entry 3). The use of KOH

afforded the product in 25% yield (Entry 4). When pre-prepared dimethylsulfoxonium methylide was used, 9 the α , β -unsaturated nitrone **1a** remained in the reaction mixture for a longer time than in the reactions via the *in situ* ylide generation. However, the final yield was not improved (Entry 5). When the temperature was slightly decreased, the dihydrooxazine **2a** was obtained in improved chemical yield of 43% yield (Entry 6). Monitoring the reactions by TLC suggested that various by-products, whose structures were not confirmed, were gradually produced. One reason for this might be the decomposition of dimethylsulfoxonium methylide to generate a reactive carbene, which caused undesired reactions with the organic materials. To prevent this interference, several alkenes were examined as scavengers to consume the methylene carbene as it was formed. When 2-methyl-2-butene was employed, the chemical yield was improved up to 54% yield (Entry 7). The reaction proceeded well even at 25 °C, albeit with slight decrease of the yield (Entry 8).

Table 1. Reaction of Sulfur Ylides with α,β -Unsaturated Nitrone 1a

Ph (CH₃)₃S(O)_nI base (3 equiv) THF,
$$T \circ C$$
, time Ph 2a

Entry	Sulfur ylide	T/°C	time	Yield/%	Recovery of 1a/% ^a
1	(CH ₃) ₃ SI / NaH	65	5 h		59
2	$(CH_3)_3S(O)I / NaH$	65	6 h	33	3
3	(CH ₃) ₃ S(O)I / NaOH	65	2 h	4	82
4	$(CH_3)_3S(O)I / KOH$	65	6 h	25	5
5	$\mathrm{CH_2S}(\mathrm{O})(\mathrm{CH_3})_2^{\ b}$	65	22 h	36	2
6		40	2 d	43	3
7^c		40	2 d	54	2
8 ^c		25	3 d	49	1

^aRecovery of **1a** was measured by ¹H NMR of the crude products using CH₂Br₂ as an internal standard. ^bPre-prepared dimethylsulfoxonium methylide was used. ^c10 Equiv of 2-methyl-2-butene was added.

With the optimal reaction conditions in hand, we examined the substrate scope of the reaction as shown in Table 2. A series of β -tolyl-substituted α,β -unsaturated nitrones **1b-d** were tolerated (Entries 2-4). Notably, β -mesityl-substituted α,β -unsaturated nitrone **1e** underwent the present

addition-rearrangement transformation to afford 3,6-dihydro-2H-1,2-oxazine 2e in 70% yield (Entry 5). β -(4-Methoxyphenyl) nitrone 1f gave 2f in 56% yield (Entry 6). β -(Di- and trimethoxyphenyl) substituted nitrones 1g-j were tolerated to furnish the products 2g-j in moderate yields (Entries 7–10). The structure of 2j was unambiguously determined by X-ray crystallographic analysis (see Experimental section). N,N-Dimethylamino and chloro groups could be introduced to the aromatic ring of the nitrones to give 3,6-dihydro-2H-1,2-oxazines 2k and 2l in about 40% yields (Entries 11 and 12). In the case of nitrone 1m with aliphatic substituents, the chemical yield was rather low (Entry 13). An α -methyl-substituted nitrone 1n could be a substrate in the present transformation (Entry 14). Although the reaction of 1n with a benzyl substituent on nitrogen produced several byproducts which were not observed in the case of N-methyl substituted nitrone 1n the dihydrooxazine 2n0 was obtained in 2n2 yield (Entry 2n5).

Table 2. Reaction of Dimethylsulfoxonium Methylide with α,β-Unsaturated Nitrones

	ı				
Entry	R^1	R^2	R^3		Yield/%
1	Ph	Н	Me	a	54
2	2-MeC_6H_4			b	50
3	$3\text{-MeC}_6\text{H}_4$			c	45
4	$4\text{-MeC}_6\text{H}_4$			d	61
5 ^a	$2,4,6-Me_3C_6H_2$			e	70
6	4-(MeO)C ₆ H ₄			f	56
7	$2,3-(MeO)_2C_6H_3$			g	52
8	$2,4-(MeO)_2C_6H_3$			h	47
9	2,5-(MeO) ₂ C ₆ H ₃			i	43
10^a	2,4,6-(MeO) ₃ C ₆ H ₂			j	55
11 ^a	$4-(Me_2N)C_6H_4$			k	40
12	$4-ClC_6H_4$			l	42
13	$Ph(CH_2)_2$			m	14
14	Ph	Me	Me	n	54
15 ^b	$2,4,6-Me_3C_6H_2$	Н	PhCH ₂	0	42

^aReaction was carried out at 60 °C. ^bReaction was carried out at 25 °C using 10 equiv of dimethylsulfoxonium methylide.

In order to gain insight into the present transformation, the nitrone 1a was treated with deuterated dimethylsulfoxonium methylide generated *in situ* from trimethylsulfoxonium- d_9 iodide and NaH in THF. It was confirmed that deuterated methylene was incorporated at the C_3 position by IH NMR analysis of the obtained 3,6-dihydro-2H-1,2-oxazine 2a- d_2 (eq 1). Based on this result, the present reaction was proposed to proceed via 1,3-addition of dimethylsulfoxonium methylide to the nitrone 1a followed by ring closure by nitrogen to produce aziridine N-oxide I. The subsequent rearrangement (Meisenheimer rearrangement) produced 3,6-dihydro-2H-1,2-oxazine 2a as shown

in Scheme 2.

Scheme 2. Plausible Reaction Mechanism for Sulfur Ylide-Mediated Formation of 3,6-Dihydro-2*H*-1,2-oxazine 2a

$$\begin{array}{c} \overline{O} \overset{+}{\text{Me}} & \overline{O} \overset{\text{Me}}{\text{N}} & \overline{O} \\ + \overline{C} H_2 - \overset{+}{\text{S}} (O) \text{Me}_2 & \overline{O} & \overline{C} H_2 \\ \hline & 1a & \overline{O} & \overline{O} & \overline{C} H_2 \\ \hline & \overline{O} & \overline{O} & \overline{C} H_2 \\ \hline & \overline{O} & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 & \overline{C} H_2 \\ \hline & \overline{C} H_2 \\ \hline & \overline{C} H_2 \\ \hline & \overline{C} H_2 \\ \hline & \overline{C} H_2 \\ \hline & \overline{C} H_2 &$$

In conclusion, we developed the unique synthesis of 3,6-dihydro-2H-1,2-oxazines via 1,3-addition of dimethylsulfoxonium methylide to α,β -unsaturated nitrones. 3,6-Dihydro-2H-1,2-oxazines, often obtained by hetero Diels-Alder reactions of nitroso compounds with 1,3-dienes, are useful heterocycles especially for the synthesis of biologically active nitrogen-containing compounds. α,β -Unsaturated nitrones could be readily available from α,β -unsaturated aldehydes and are slightly more stable than 1,3-dienes. The present method provides a useful, unique, and alternative way to provide 3,6-dihydro-2H-1,2-oxazines.

EXPERIMENTAL SECTION

General Method. ¹H NMR spectra were recorded on a 400 MHz NMR spectrometer. Chemical shifts δ are reported in ppm using TMS as an internal standard. Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, quint = quintet, m = multiplet), coupling constant (*J*) and integration. ¹³C NMR spectra were recorded on a 100 MHz NMR spectrometer. The chemical shifts are reported relative to CDCl₃ (δ = 77.0 ppm). The wavenumbers of maximum absorption peaks in IR spectra are presented in cm⁻¹. All of the melting points were measured with a micro melting point apparatus. HRMS (ESI and FAB) spectra were measured with

quadrupole and TOF mass spectrometers. Dehydrated solvents were purchased for the reactions and used without further desiccation.

(*Z*)-*N*-((*E*)-3-Phenylallylidene)methanamine oxide (1a): 18,19 To a stirred mixture of (*E*)-cinnamaldehyde (3.96 g, 30 mmol), triethylamine (5.58 mL, 40 mmol) and anhydrous Na₂SO₄ (14.2 g, 100 mmol) in dichloromethane (60 mL) was added *N*-methylhydroxylamine hydrochloride (1.67 g, 20 mmol) at 0 °C. After stirring for 2 d at rt, the reaction mixture was partitioned between water and CHCl₃. The aqueous layer was extracted with CHCl₃. The combined organic layers were dried over anhydrous Na₂SO₄, filtered and evaporated under reduced pressure. The residue was purified by silica gel column chromatography (AcOEt/MeOH = 6/1) to give the nitrone **1a** (2.06 g, 64% yield) as a solid. Mp. 86–87 °C. 1 H NMR (CDCl₃, 400 MHz): δ 7.52 (d, 2H, J = 6.9 Hz), 7.45 (dd, 1H, J = 16.5, 9.6 Hz), 7.38–7.29 (m, 3H), 7.25 (d, 1H, J = 9.6 Hz), 6.96 (d, 1H, J = 16.5 Hz), 3.77 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 138.1, 137.6, 135.9, 129.2, 128.8, 127.3, 118.4, 52.4. IR (KBr) 3056, 1591, 1403, 1191, 1139, 980, 947, 750, 697 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₀H₁₂NO 162.0919; Found 162.0922.

In a similar manner, nitrones **1b–1o** were obtained from the corresponding α,β -unsaturated aldehydes.

(*Z*)-*N*-((*E*)-3-(*o*-Tolyl)allylidene)methanamine oxide (1b): The compound 1b (2.00 g, 68% yield, 17 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). Mp. 85–86 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.69–7.66 (m, 1H), 7.38 (dd, 1H, J = 16.0, 9.6 Hz), 7.28 (d, 1H, J = 9.6 Hz), 7.23–7.15 (m, 4H), 3.76 (s, 3H), 2.37 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 137.8, 136.4, 135.3, 134.6, 130.6, 129.0, 126.3, 125.6, 119.0, 52.3, 19.7. IR (KBr) 3043, 1561, 1407, 1184, 1146, 971, 955, 746 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₁H₁₄NO 176.1075; Found 176.1074.

(*Z*)-*N*-((*E*)-3-(*m*-Tolyl)allylidene)methanamine oxide (1c): The compound 1c (2.95 g, 99% yield, 16 mmol scale) was obtained as an oil after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). ¹H NMR (CDCl₃, 400 MHz): δ 7.43 (dd, 1H, J = 16.0, 9.6 Hz), 7.36 (s, 1H), 7.31 (d, 1H, J = 7.8 Hz), 7.26–7.22 (m, 2H), 7.13 (d, 1H, J = 7.3 Hz), 6.93 (d, 1H, J = 16.0 Hz), 3.76 (s, 3H), 2.35 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 138.4, 138.3, 137.7, 135.9, 130.1, 128.7, 127.8, 124.6, 118.2, 52.3, 21.3. IR (neat) 2943, 1560, 1484, 1395, 957, 778, 691 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₁H₁₄NO 176.1075; Found 176.1076.

- (*Z*)-*N*-((*E*)-3-(*p*-Tolyl)allylidene)methanamine oxide (1d): 19,20 The compound 1d (0.11 g, 60% yield, 1.0 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). Mp. 121–123 °C. 1 H NMR (CDCl₃, 400 MHz): δ 7.42–7.36 (m, 1H), 7.41 (d, 2H, J = 7.8 Hz), 7.22 (d, 1H, J = 9.6 Hz), 7.15 (d, 2H, J = 7.8 Hz), 6.92 (d, 1H, J = 16.0 Hz), 3.74 (s, 3H), 2.34 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 139.4, 138.2, 137.8, 133.1, 129.5, 127.2, 117.3, 52.1, 21.3. IR (KBr) 3013, 1553, 1510, 1392, 1174, 1144, 982, 789 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₁H₁₄NO 176.1075; Found 176.1072.
- (*Z*)-*N*-((*E*)-3-Mesitylallylidene)methanamine oxide (1e): The compound 1e (2.25 g, 73% yield, 15.2 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). Mp. 130–131 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.26 (d, 1H, J = 8.2 Hz), 7.09 (d, 1H, J = 16.9 Hz), 7.02 (dd, 1H, J = 16.9, 8.2 Hz), 6.88 (s, 2H), 3.76 (s, 3H), 2.34 (s, 6H), 2.27 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 138.1, 137.7, 136.5, 131.92, 131.85, 129.2, 123.3, 52.3, 21.3, 21.0. IR (KBr) 3002, 1574, 1398, 1194, 1141, 979, 855, 724 cm⁻¹. HRMS (ESI) m/z: [M+Na]⁺ Calcd for C₁₃H₁₇NONa 226.1208; Found 226.1223.
- (*Z*)-*N*-((*E*)-3-(4-Methoxyphenyl)allylidene)methanamine oxide (1f): ^{19,20} The compound 1f (1.46 g, 51% yield, 15.0 mmol scale) was obtained as a solid by recrystalization (AcOEt/hexane). Mp. 101–102 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.47 (d, 2H, J = 8.7 Hz), 7.31 (dd, 1H, J = 16.0, 9.6 Hz), 7.24 (d, 1H, J = 9.6 Hz), 6.93 (d, 1H, J = 16.0 Hz), 6.88 (d, 2H, J = 8.7 Hz), 3.83 (s, 3H), 3.76 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 160.9, 138.7, 138.2, 129.0, 128.6, 116.0, 114.3, 55.4, 52.0. IR (KBr) 2950, 1599, 1510, 1401, 1301, 1257, 1172, 1140, 1026, 974, 946, 812 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₁H₁₄NO₂ 192.1025; Found 192.1029.
- (*Z*)-*N*-((*E*)-3-(2,3-Dimethoxyphenyl)allylidene)methanamine oxide (1g): The compound 1g (1.23 g, 34% yield, 16.6 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). Mp. 85–86 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.45 (dd, 1H, J = 16.0, 9.6 Hz), 7.33 (d, 1H, J = 16.0 Hz), 7.32–7.29 (m, 2H), 7.06 (t, 1H, J = 8.0 Hz), 6.89 (d, 1H, J = 7.8 Hz), 3.88 (s, 3H), 3.83 (s, 3H), 3.77 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 152.9, 147.3, 138.2, 132.3, 130.1, 124.3, 119.2, 118.1, 112.8, 61.3, 55.8, 52.3. IR (KBr) 2940, 1557, 1476, 1387, 1268, 1178, 1146, 1061, 998, 983, 744 cm⁻¹. HRMS (ESI) m/z: [M+Na]⁺ Calcd for C₁₂H₁₅NO₃Na 244.0950; Found 244.0963.

- (*Z*)-*N*-((*E*)-3-(2,4-Dimethoxyphenyl)allylidene)methanamine oxide (1h): The compound 1h (2.29 g, 65% yield, 15.9 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 3/1). Mp. 99–100 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.58 (d, 1H, J = 8.2 Hz), 7.35 (dd, 1H, J = 15.6, 9.6 Hz), 7.27–7.20 (m, 2H), 6.51 (d, 1H, J = 8.2 Hz), 6.43 (s, 1H), 3.85 (s, 3H), 3.83 (s, 3H), 3.73 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 161.9, 158.6, 138.9, 133.0, 128.2, 118.1, 116.3, 105.3, 98.3, 55.5, 55.4, 52.0. IR (KBr) 3004, 1594, 1505, 1450, 1391, 1315, 1274, 1297, 1181, 1027, 945, 828 cm⁻¹. HRMS (FAB⁺) m/z: [M+H]⁺ Calcd for C₁₂H₁₆NO₃ 222.1130; Found 222.1131.
- (*Z*)-*N*-((*E*)-3-(2,5-Dimethoxyphenyl)allylidene)methanamine oxide (1i): The compound 1i (2.33 g, 81% yield, 13.0 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). Mp. 68–70 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.43 (dd, 1H, J = 16.5, 9.6 Hz), 7.32 (d, 1H, J = 16.5 Hz), 7.27 (d, 1H, J = 9.6 Hz), 7.17 (d, 1H, J = 2.8 Hz), 6.86 (dd, 1H, J = 9.2, 2.8 Hz), 6.82 (d, 1H, J = 9.2 Hz), 3.83 (s, 3H), 3.80 (s, 3H), 3.76 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 153.7, 151.8, 138.5, 132.7, 125.4, 118.4, 116.9, 112.4, 110.6, 56.1, 55.8, 52.3. IR (KBr) 2943, 1563, 1496, 1413, 1281, 1221, 1188, 1039, 969, 824, 801, 707 cm⁻¹. HRMS (ESI) m/z: [M+Na]⁺ Calcd for C₁₂H₁₅NO₃Na 244.0950; Found 244.0965.
- (*Z*)-*N*-((*E*)-3-(2,4,6-Trimethoxyphenyl)allylidene)methanamine oxide (1j): The compound 1j (0.94 g, 53% yield, 7.1 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1, 3/1). Mp. 147–148 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.69 (dd, 1H, J = 16.5, 10.1 Hz), 7.26 (d, 1H, J = 16.5 Hz), 7.17 (d, 1H, J = 10.1 Hz), 6.11 (s, 2H), 3.87 (s, 6H), 3.84 (s, 3H), 3.71 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 161.8, 160.3, 140.8, 130.0, 119.2, 107.2, 90.3, 55.7, 55.3, 51.9. IR (KBr) 2943, 1593, 1463, 1408, 1319, 1200, 1119, 1038, 976, 949, 822 cm⁻¹. HRMS (ESI) m/z: [M+Na]⁺ Calcd for C₁₃H₁₇NO₄Na 274.1055; Found 274.1074.
- (*Z*)-*N*-((*E*)-3-(4-(Dimethylamino)phenyl)allylidene)methanamine oxide (1k): The compound 1k (0.82 g, 86% yield, 4.7 mmol scale) was obtained as a solid by recrystalization (MeOH/hexane). Mp. 105–106 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.42 (d, 2H, J = 8.7 Hz), 7.27–7.16 (m, 2H), 6.87 (d, 1H, J = 15.6 Hz), 6.66 (d, 2H, J = 8.7 Hz), 3.73 (s, 3H), 3.01 (s, 6H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 151.0, 139.3, 138.8, 128.9, 123.9, 113.7, 111.9, 51.8, 40.1. IR (KBr) 1598, 1325, 1371, 1173, 809 cm⁻¹. HRMS (ESI) m/z: [M+Na]⁺ Calcd for C₁₂H₁₆N₂ONa 227.1160; Found 227.1173.

- (*Z*)-*N*-((*E*)-3-(4-Chlorophenyl)allylidene)methanamine oxide (11): 20,21 The compound 11 (1.26 g, 77% yield, 8.4 mmol scale) was obtained as a solid after purification by silica gel column chromatography (AcOEt/MeOH = 6/1). Mp. 130–132 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.44 (d, 2H, J = 8.7 Hz), 7.40 (dd, 1H, J = 16.0, 10.1 Hz), 7.32 (d, 2H, J = 8.7 Hz), 7.25 (d, 1H, J = 10.1 Hz), 6.93 (d, 1H, J = 16.0 Hz), 3.77 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 137.6, 136.8, 135.0, 134.5, 129.1, 128.4, 118.8, 52.4. IR (KBr) 3034, 1557, 1488, 1391, 1186, 1140, 972, 956, 803 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₀H₁₁NOCl 196.0529; Found 196.0524.
- (*Z*)-*N*-((*E*)-5-Phenylpent-2-en-1-ylidene)methanamine oxide (1m): The compound 1m (0.59 g, 18% yield, 17.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (AcOEt/MeOH = 20/1, 10/1, 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.32–7.13 (m, 5H), 7.03 (d, 1H, J = 9.6 Hz), 6.80 (dd, 1H, J = 16.0, 9.6 Hz), 6.24 (dt, 1H, J = 16.0, 6.9 Hz), 3.68 (s, 3H), 2.77 (t, 2H, J = 7.3 Hz), 2.55 (td, 2H, J = 7.3, 6.9 Hz). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 141.9, 140.9, 137.4, 128.4, 128.3, 126.1, 121.4, 52.0, 34.89, 34.86. IR (neat) 3026, 2926, 1631, 1603, 1571, 1454, 1407, 1149, 977, 748, 701 cm $^{-1}$; HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₂H₁₆NO 190.1232; Found 190.1237.
- (*Z*)-*N*-((*E*)-2-Methyl-3-phenylallylidene)methanamine oxide (1n):²² The compound 1n (1.40 g, 53% yield, 15.0 mmol scale) was obtained as an oil after purification by silica gel column chromatography (AcOEt/MeOH = 10/1, 6/1). ¹H NMR (CDCl₃, 400 MHz): δ 8.26 (s, 1H), 7.36–7.19 (m, 5H), 6.93 (s, 1H), 3.78 (s, 3H), 2.18 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 137.9, 137.0, 134.5, 129.5, 128.1, 127.7, 127.4, 54.4, 17.7. IR (KBr) 3056, 1587, 1558, 1167, 957, 751, 699 cm⁻¹. HRMS (ESI) *m/z*: [M+H]⁺ Calcd for C₁₁H₁₄NO 176.1075; Found 176.1074.
- (*Z*)-*N*-((*E*)-3-Mesitylallylidene)-1-phenylmethanamine oxide (1o): The compound 1o (0.59 g, 88% yield, 2.4 mmol scale) was obtained as a solid after purification by silica gel column chromatography (hexane/AcOEt = 2/1, 1/1, AcOEt only, then AcOEt/MeOH = 10/1). Mp. 180–182 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.49–7.39 (m, 5H), 7.24–7.21 (m, 1H), 7.06–7.04 (m, 2H), 6.86 (s, 2H), 4.95 (s, 2H), 2.32 (s, 6H), 2.26 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 137.7, 136.9, 136.8, 136.6, 132.9, 131.9, 129.4, 129.2, 128.9, 123.2, 69.2, 21.4, 21.0; one signal overlaps. IR (KBr) 3061, 1608, 1556, 1433, 1331, 1205, 1185, 1119, 984, 938, 864, 765, 701 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₉H₂₂NO 280.1701; Found 280.1699.

Representative Procedure for Synthesis of 3,6-Dihydro-2*H*-1,2-oxazines for 2d.

Under an Ar atmosphere, a mixture of α,β -unsaturated nitrone **1d** (175 mg, 1.0 mmol), dimethylsulfoxonium methylide⁹ (4.8 mL, 0.63 M in THF, 3.0 mmol), and 2-methyl-2-butene (1.1 mL, 10 mmol) in THF (1.2 mL) was stirred at 40 °C (oil bath temperature) for 2 d. After cooling to rt, brine was added and extracted with CHCl₃. Combined organic layers were dried over Na₂SO₄ and concentrated. The residue was purified by column chromatography on silica gel (hexane/AcOEt = 15/1, 10/1, 6/1) to give **2d** (115 mg, 61% yield).

In a similar manner, syntheses of 3,6-dihydro-2*H*-1,2-oxazines 2a–2c, and 2e–2o were carried out.

- **2-Methyl-6-phenyl-3,6-dihydro-2***H***-1,2-oxazine (2a):** The compound **2a** (47 mg, 54% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.40–7.28 (m, 5H), 6.02–5.87 (m, 2H), 5.45 (brs, 1H), 3.30 (brs, 2H), 2.71 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 139.2, 128.8, 128.4, 128.1, 127.8, 124.1, 79.5, 56.3, 46.1. IR (neat) 3033, 2953, 2919, 2852, 2805, 1656, 1493, 1454, 1436, 1274, 1097, 1056, 994, 957, 924, 884, 792, 753, 700, 680 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₁H₁₄NO 176.1075; Found 176.1080.
- **2-Methyl-6-(***o***-tolyl)-3,6-dihydro-2***H***-1,2-oxazine (2b):** The compound **2b** (47 mg, 50% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.33 (d, 1H, J = 6.4 Hz), 7.24–7.06 (m, 3H), 6.05–5.97 (m, 1H), 5.93 (d, 1H, J = 10.1 Hz), 5.71 (br, 1H), 3.31 (brs, 2H), 2.71 (s, 3H), 2.43 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 150.0, 136.9, 130.5, 128.6, 128.2, 127.8, 125.9, 124.3, 76.6, 56.3, 46.1, 19.0. IR (neat) 2953, 2853, 2804, 1652, 1490, 1462, 1437, 1096, 1058, 995, 952, 888, 751, 670 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₂H₁₆NO 190.1232; Found 190.1233.
- **2-Methyl-6-(***m***-tolyl)-3,6-dihydro-2***H***-1,2-oxazine (2c):** The compound **2c** (43 mg, 45% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.30–7.10 (m, 4H), 6.06–5.86 (m, 2H), 5.43 (br, 1H), 3.31 (brs, 2H), 2.72 (s, 3H), 2.35 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 150.0, 138.1, 137.6, 129.0, 128.5, 128.3, 124.9, 124.0, 79.8, 56.3, 46.1, 21.4. IR (neat) 2953, 2920, 2853, 2805, 1657, 1608, 1489, 1461, 1437, 1098, 1058, 994, 967, 853, 785, 761, 702, 677 cm $^{-1}$. HRMS (ESI) m/z Calcd for $C_{12}H_{16}NO$ [M+H] $^{+}$ 190.1232; Found 190.1231.
- **2-Methyl-6-(***p***-tolyl)-3,6-dihydro-2***H***-1,2-oxazine (2d):** The compound **2d** (115 mg, 61% yield, 1.0 mmol scale) was obtained as an oil after purification by silica gel column chromatography

(hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.27 (d, 2H, J = 8.1 Hz), 7.15 (d, 2H, J = 8.1 Hz), 6.00–5.83 (m, 2H), 5.42 (br, 1H), 3.29 (brs, 2H), 2.70 (s, 3H), 2.33 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 138.0, 136.1, 129.1, 127.9, 124.1, 79.5, 56.3, 46.1, 21.2. IR (neat) 2953, 2920, 2852, 1654, 1615, 1514, 1436, 1097, 1057, 994, 957, 888, 813, 659 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₂H₁₆NO 190.1232; Found 190.1227.

6-Mesityl-2-methyl-3,6-dihydro-2*H***-1,2-oxazine (2e):** The compound **2e** (76 mg, 70% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 6.74 (s, 2H), 5.92–5.81 (m, 2H), 5.74 (d, 1H, J = 10.1 Hz), 3.40–3.18 (m, 2H), 2.66 (s, 3H), 2.30 (s, 6H), 2.15 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 137.6, 137.5, 131.4, 129.8, 128.9, 123.1, 76.2, 56.1, 46.2, 20.7, 20.6. IR (neat) 2952, 2918, 2852, 1658, 1611, 1459, 1435, 994, 869, 797 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₄H₂₀NO 218.1545; Found 218.1548.

6-(4-Methoxyphenyl)-2-methyl-3,6-dihydro-2*H***-1,2-oxazine (2f):** The compound **2f** (57 mg, 56% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.30 (d, 2H, J = 8.2 Hz), 6.87 (d, 2H, J = 8.2 Hz), 6.02–5.95 (m, 1H), 5.90 (d, 1H, J = 10.1 Hz), 5.41 (br, 1H), 3.79 (s, 3H), 3.29 (brs, 2H), 2.70 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 159.5, 131.3, 129.3, 128.9, 124.1, 113.8, 79.1, 56.3, 55.2, 46.1. IR (neat) 2953, 2836, 1653, 1611, 1513, 1462, 1248, 1173, 1096, 1056, 1035, 995, 887, 833, 660 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₂H₁₆NO₂ 206.1181; Found 206.1178.

6-(2,3-Dimethoxyphenyl)-2-methyl-3,6-dihydro-2*H***-1,2-oxazine (2g):** The compound **2g** (61 mg, 52% yield. 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 3/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.06–6.95 (m, 2H), 6.87 (d, 1H, J = 7.8 Hz), 5.99–5.81 (m, 3H), 3.88 (s, 3H), 3.86 (s, 3H), 3.30 (brs, 2H), 2.71 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): 152.6, 147.1, 133.1, 128.9, 124.0, 123.8, 120.2, 112.1, 73.4, 61.3, 56.3, 55.7, 46.1. IR (neat) 2936, 2834, 1657, 1586, 1480, 1431, 1277, 1055, 1008, 858, 574, 665 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₃H₁₈NO₃ 236.1287; Found 236.1292.

6-(2,4-Dimethoxyphenyl)-2-methyl-3,6-dihydro-2*H***-1,2-oxazine (2h):** The compound **2h** (56 mg, 47% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 3/1). ¹H NMR (CDCl₃, 400 MHz): δ 7.30–7.25 (m, 1H), 6.48–6.45 (m, 1H), 6.45 (s, 1H), 5.97–5.83 (m, 3H), 3.81 (s, 3H), 3.79 (s, 3H), 3.28 (brs, 2H), 2.70 (s,

3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 160.6, 158.1, 129.4, 123.5, 120.0, 104.2, 98.4, 72.6, 56.4, 55.5, 55.3, 46.1. IR (neat) 2953, 2834, 1613, 1588, 1506, 1464, 1288, 1208, 1157, 1038, 832, 665 cm $^{-1}$. HRMS (FAB⁺) m/z: [M+H]⁺ Calcd for C₁₃H₁₈NO₃ 236.1287; Found 236.1280.

6-(2,5-Dimethoxyphenyl)-2-methyl-3,6-dihydro-2*H***-1,2-oxazine (2i):** The compound **2i** (51 mg, 43% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 3/1). ¹H NMR (CDCl₃, 400 MHz): δ 7.03–6.93 (m, 1H), 6.81 (d, 1H, J = 9.2 Hz), 6.78 (dd, 1H, J = 9.2, 2.8 Hz), 5.96–5.81 (m, 3H), 3.80 (s, 3H), 3.76 (s, 3H), 3.29 (brs, 2H), 2.73 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 153.5, 151.0, 138.3, 129.0, 123.5, 114.1, 113.3, 111.9, 73.1, 56.3, 56.2, 55.6, 46.1. IR (neat) 2952, 2839, 1591, 1497, 1278, 1219, 1048, 806, 715 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₃H₁₈NO₃ 236.1287; Found 236.1282.

6-(2,4,6-Trimethoxyphenyl)-2-methyl-3,6-dihydro-2*H***-1,2-oxazine (2j):** The compound **2j** (73 mg, 55% yield, 0.5 mmol scale) was obtained as a solid after purification by silica gel column chromatography (hexane/AcOEt = 3/1). Mp 113–114 °C. ¹H NMR (CDCl₃, 400 MHz): δ 6.02 (s, 2H), 5.98 (brs, 1H), 5.72 (brs, 2H), 3.71 (s, 3H), 3.69 (s, 6H), 3.22 (brs, 2H), 2.65 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 161.4, 160.5, 130.2, 120.6, 107.6, 91.3, 71.0, 56.2, 55.9, 55.2, 46.2. IR (KBr) 2909, 2800, 1606, 1459, 1204, 1125, 957, 806 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₄H₂₀NO₄ 266.1392; Found 266.1394.

Recrystallization from hexane/AcOEt gave the crystal **2j**. Crystal data: $C_{14}H_{19}NO_4$, $M_r = 265.30$, monoclinic, $P2_1/c$, a = 10.5750(5), b = 8.7937(4), c = 14.6350(7) Å, $\beta = 91.4380(10)$. V = 1360.53(11) Å³, Z = 4, $D_{calcd} = 1.295$ g cm⁻³, R = 0.0419 ($R_w = 0.1726$) for 2658 reflections. CCDC 1991113 (**2j**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

N,*N*-Dimethyl-4-(2-methyl-3,6-dihydro-2*H*-1,2-oxazin-6-yl)aniline (2k): The compound 2k (44 mg, 40% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 3/1). ¹H NMR (CDCl₃, 400 MHz): δ 7.24 (d, 2H, J = 8.7 Hz), 6.70 (d, 2H, J = 8.7 Hz), 6.02–5.84 (m, 2H), 5.38 (br, 1H), 3.28 (brs, 2H), 2.93 (s, 6H), 2.69 (s, 3H). ¹³C{ ¹H} NMR (CDCl₃, 100 MHz): 150.7, 129.4, 129.1, 126.8, 123.9, 112.4, 79.5, 56.3, 46.1, 40.6. IR (neat) 2850, 2802, 1614, 1523, 1351, 1165, 1053, 812 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₃H₁₉N₂O 219.1497; Found 219.1494.

- **6-(4-Chlorophenyl)-2-methyl-3,6-dihydro-2***H***-1,2-oxazine (2l):** The compound **2l** (44 mg, 42% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.32 (s, 4H), 6.01 (d, 1H, J = 9.6 Hz), 5.96–5.84 (m, 1H), 5.41 (br, 1H), 3.30 (brs, 2H), 2.70 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 149.8, 137.8, 133.8, 129.2, 128.6, 124.6, 78.9, 56.2, 46.1. IR (neat) 2954, 2922, 2852, 1653, 1596, 1491, 1436, 1408, 1260, 1092, 1059, 1015, 995, 887, 807, 720 cm $^{-1}$. HRMS (ESI) m/z: [M+H] $^{+}$ Calcd for C₁₁H₁₃NOCl 210.0686; Found 210.0685.
- **2-Methyl-6-phenethyl-3,6-dihydro-2***H***-1,2-oxazine (2m):** The compound **2m** (14 mg, 14% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). ¹H NMR (CDCl₃, 400 MHz): δ 7.30–7.07 (m, 5H), 5.87–5.81 (m, 1H), 5.75 (d, 1H, J = 10.1 Hz), 4.45 (br, 1H), 3.25–3.14 (m, 2H), 2.82–2.67 (m, 2H), 2.70 (s, 3H), 1.84–1.72 (m, 2H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): 142.0, 129.4, 128.5, 128.3, 125.7, 123.7, 76.1, 56.4, 46.0, 35.3, 31.5. IR (neat) 3027, 2922, 2852, 1604, 1496, 1455, 1436, 1103, 1055, 993, 750, 699 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₃H₁₈NO 204.1388; Found 204.1389.
- **2,5-Dimethyl-6-phenyl-3,6-dihydro-2***H***-1,2-oxazine (2n):** The compound **2n** (51 mg, 54% yield, 0.5 mmol scale) was obtained as an oil after purification by silica gel column chromatography (hexane/AcOEt = 6/1). ¹H NMR (CDCl₃, 400 MHz): δ 7.35–7.28 (m, 5H), 5.71 (s, 1H), 5.25 (br, 1H), 3.38–3.20 (m, 2H), 2.67 (s, 3H), 1.45 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 138.4, 135.1, 128.7, 128.33, 128.26, 119.6, 83.4, 56.6, 46.0, 18.7. IR (neat) 2916, 2852, 2801, 1493, 1454, 1437, 1379, 1270, 1087, 1039, 995, 942, 900, 850, 755, 701 cm⁻¹. HRMS (ESI) m/z: [M+H]⁺ Calcd for C₁₂H₁₆NO 190.1232; Found 190.1235.
- **2-Benzyl-6-mesityl-3,6-dihydro-2***H***-1,2-oxazine (2o):** The compound **2o** (16 mg, 42% yield, 0.13 mmol scale) was obtained as an oil after purification by preparative TLC (hexane/AcOEt = 10/1). 1 H NMR (CDCl₃, 400 MHz): δ 7.38 (d, 2H, J = 6.9 Hz), 7.32–7.20 (m, 3H), 6.79 (s, 2H), 5.95 (brs, 1H), 5.94–5.81 (m, 2H), 4.12 (d, 1H, J = 13.7 Hz), 3.87 (d, 1H, J = 13.7 Hz), 3.45–3.31 (m, 2H), 2.34 (s, 6H), 2.22 (s, 3H). 13 C{ 1 H} NMR (CDCl₃, 100 MHz): δ 137.9, 137.5, 136.9, 131.6, 129.8, 129.5, 128.9, 128.7, 128.1, 128.0, 127.1, 123.2, 76.3, 62.7, 53.9, 20.8, 20.7. IR (neat) 3033, 2919, 2859, 2799, 1610, 1494, 1454, 1378, 1339, 1085, 1030, 970, 742, 698 cm $^{-1}$. HRMS (FAB $^{+}$) m/z: [M+H] $^{+}$ Calcd for C₂₀H₂₄NO 294.1858; Found 294.1853.

2-Methyl-6-phenyl-3,6-dihydro-2H-(3,3-D₂)1,2-oxazine (2a- d_2): Under an Ar atmosphere, a

mixture of trimethylsulfoxonium-d₉ iodide¹⁴ (344 mg, 1.50 mmol) and NaH (60 mg, 60% w/w

dispersion in mineral oil, 1.50 mmol) in THF (3 mL) was refluxed for 2 h. After cooling to 0 °C,

α,β-unsaturated nitrone 1a (81 mg, 0.50 mmol) in THF (2 mL) was added and the resulting mixture

was stirred for 3 h at 70 °C (oil bath temperature). After cooling to rt, sat aqueous NH₄Cl solution

was added and extracted with CHCl₃. Combined organic layers were dried over Na₂SO₄ and

concentrated. The residue was purified by column chromatography on silica gel (hexane/AcOEt =

15/1, 12/1, 8/1) to give a deuterated 3,6-dihydro-2H-1,2-oxazine 2a- d_2 as an oil (37 mg, 42% yield,

86%-D). ¹H NMR (CDCl₃, 400 MHz): δ 7.45–7.28 (m, 5H), 6.04–5.88 (m, 2H), 5.45 (brs, 1H),

3.32–3.27 (m, 0.27H), 2.72 (s, 3H). ${}^{13}C\{{}^{1}H\}$ NMR (CDCl₃, 100 MHz): δ 138.9, 128.7, 128.4, 128.2,

127.8, 124.0, 79.6, 55.7 (quint, J = 18 Hz), 46.1. IR (neat) 3033, 2953, 2185, 2059, 1493, 1454,

1274, 1095, 1044, 921, 893, 748, 699 cm⁻¹. HRMS (ESI) Calcd for C₁₁H₁₂D₂NO [M+H]⁺: 178,1201,

found 178.1201.

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ACKNOWLEDGEMENTS

This work was partially supported by Grants-in-Aid from the Japan Society for the Promotion of

Science (18K05102 and 17K05778), Kanazawa University SAKIGAKE Project, and the Kanazawa

University CHOZEN Project.

SUPPORTING INFORMATION

Copies of ¹H NMR and ¹³C NMR spectra of products, cif files.

REFERENCES AND NOTES

(1) Corey, E. J.; Chaykovsky, M. Dimethyloxosulfonium Methylide ((CH₃)₂SOCH₂) and

Dimethylsulfonium Methylide ((CH₃)₂SCH₂). Formation and Application to Organic Synthesis.

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- J. Am. Chem. Soc. 1965, 87, 1353-1364.
- (2) General reviews of sulfur ylides: (a) Trost, B. M.; Melvin, Jr. L. S. T. In Sulfur Ylide; Academic Press, Inc.: New York, 1975. (b) Gololobov, Y. G.; Nesmeyanov, A. N.; Lysenko, V. P.; Boldeskul, I. E. Twenty-five Years of Dimethylsulfoxonium Methylide (Corey's reagent). Tetrahedron 1987, 43, 2609–2651. (c) Aggarwal, V. K.; Badine, M. D.; Moorthie, V. A. Asymmetric Synthesis of Epoxides and Aziridines from Aldehydes and Imines. In Aziridines and Epoxides in Organic Synthesis; Yudin, A. K., Ed. Wiley-VCH: Weinheim, 2006, pp. 1–34. (d) Awasthi, C. Dimethylsulfoxonium Methylide (DSM): A Versatile Reagent. Synlett 2008, 1423–1424. (e) Yao, H. Dimethylsulfonium Methylide: A Versatile Reagent. Synlett 2009, 2201–2202. (f) Lu, L.-Q.; Li, T.-R.; Wang, Q.; Xiao, W.-J. Beyond sulfidecentric catalysis: recent advances in the catalytic cyclization reactions of sulfur ylides. Chem. Soc. Rev. 2017, 46, 4135–4149.
- (3) Sun, X.-L.; Tang, Y. Ylide-Initiated Michael Addition-Cyclization Reactions beyond Cyclopropanes. *Acc. Chem. Res.* **2008**, *41*, 937–948.
- (4) Examples of the formation of aziridines via 1,2-addition: (a) Morton, D.; Pearson, D.; Field, R. A.; Stockman, R. A. Corey-Chaykovsky Reaction of Chiral Sulfinyl Imines: A Convenient Procedure for the Formation of Chiral Aziridines. *Synlett* 2003, 1985–1988. (b) Fearraigh, M. P. Ó.; Matlock, J. V.; Illa, O.; McGarrigle, E. M.; Aggarwal, V. K. Synthesis of Isothiocineole and Application in Multigram-Scale Sulfur Ylide Mediated Asymmetric Epoxidation and Aziridination. *Synthesis* 2018, 50, 3337–3343.
- (5) (a) Lu, L.-Q.; Zhang, J.-J.; Li, F.; Cheng, Y.; An, J.; Chen, J.-R.; Xiao, W.-J. Tuning Electronic and Steric Effects: Highly Enantioselective [4+1] Pyrroline Annulation of Sulfur Ylides with α,β-Unsaturated Imines. Angew. Chem. Int. Ed. 2010, 49, 4495–4498. (b) Meng, X.-S.; Jiang, S.; Xu, X.-Y.; Wu, Q.-X.; Gu, Y.-C.; Shi, D.-Q. Stabilized Sulfur Ylide Mediated Cyclopropanations and Formal [4+1] Cycloadditions of 3-Acyl-2H-chromenones and Their Imines. Eur. J. Org. Chem. 2016, 4778–4781. An example using telluronium ylides: (c) Zheng, J.-C.; Liao, W.-W.; Tang, Y.; Sun, X.-L.; Dai, L.-X. The Michael Addition-Elimination of Ylides to α,β -Unsaturated Imines. Highly Stereoselective **Synthesis** of Vinylcyclopropanecarbaldehydes and Vinylcyclopropylaziridines. J. Am. Chem. Soc. 2005, 127, 12222-12223.

- (6) Lantos, I.; Flisak, J.; Liu, L.; Matsuoka, R.; Mendelson, W.; Stevenson, D.; Tubman, K.; Tucker, L.; Zhang, W.-Y.; Adams, J.; Sorenson, M.; Garigipati, R.; Erhardt, K.; Ross, S. Enantioselective Synthesis of 5-LO Inhibitor Hydroxyureas. Tandem Nucleophilic Addition-Intramolecular Cyclization of Chiral Nitrones. *J. Org. Chem.* 1997, 62, 5385–5391.
- (7) Soeta, T.; Ohgai, T.; Sakai, T.; Fujinami, S.; Ukaji, Y. Ring Enlargement Reaction of *C,N*-Cyclic-*N'*-acyl Azomethine Imines with Sulfonium Ylide: An Efficient Synthesis of 3-Benzazepine Derivatives. *Org. Lett.* **2014**, *16*, 4854—4857.
- (8) ¹H NMR: (a) Riddell, F. G.; Labaziewicz, H. 1,2-Oxazine Chemistry V. Vicinal, Allylic and Homoallylic Coupling Constants in the 3,6-Dihydro-1,2-oxazine Ring. *Org. Magn. Res.* 1974, 6, 599–600. ¹³C NMR: (b) Labaziewicz, H.; Riddell, F. G.; Sayer, B. G. 1,2-Oxazine Chemistry. Part 6. Conformational Analysis of Cyclohexene and a Heterocyclic Analogue by ¹³C Nuclear Magnetic Resonance Spectroscopy. *J. Chem. Soc. Perkin Trans.* 2 1977, 619–622.
- (9) (a) Sone, T.; Yamaguchi, A.; Matsunaga, S.; Shibasaki, M. Catalytic Asymmetric Synthesis of 2,2-Disubstituted Terminal Epoxides via Dimethyloxosulfonium Methylide Addition to Ketones. J. Am. Chem. Soc. 2008, 130, 10078–10079. (b) Zhang, H.; Alkayal, N.; Gnanou, Y.; Hadjichristidis, N. Anionic polymerization and polyhomologation: an ideal combination to synthesize polyethylene-based block copolymers. Chem. Commun. 2013, 49, 8952–8954.
- (10) (a) Franzen, V.; Schmidt, H. -J.; Mertz, C. Untersuchungen über Carbene, IX Carbene aus Sulfoniumsalzen. *Chem. Ber.* 1961, 94, 2942–2950. (b) Trost, B. M. Decomposition of Sulfur Ylides. Evidence for Carbene Intermediates. *J. Am. Chem. Soc.* 1966, 88, 1587–1588. (c) Kunieda, T.; Witkop, B. Preparation and Photochemistry of Pyrimidine Nucleoside Sulfonium Ylides. *J. Am. Chem. Soc.* 1971, 93, 3487–3493.
- (11) When the reactions were carried out in the presence of 5 equiv of cyclohexene, 2-methyl-2-butene, and 2,3-dimethyl-2-butene, **2a** was obtained in slightly enhanced chemical yield (46%) in all three cases, respectively.
- (12) 2-Methyl-2-butene is known as the chlorine scavenger in NaClO oxidation: Kraus, G. A.; Taschner, M. J. Model Studies for the Synthesis of Quassinoids. 1. Construction of the BCE Ring System. J. Org. Chem. 1980, 45, 1175–1176.
- (13) When (4-methylpent-3-en-1-yl)benzene (3 equiv) was used as an additive (40 °C, 2 d), **2a** was obtained in 48% yield. Unexpectedly, the corresponding cyclopropane, (2-(2,2-

- dimethylcyclopropyl)ethyl)benzene, was not observed in the crude products by ¹H NMR analysis. At present, the role of the olefinic additives was not clear yet. The ¹H NMR data of the expected cyclopropane was reported: Charette, A. B.; Molinaro, C.; Brochu, C. Stability, Reactivity, Solution, and Solid-State Structure of Halomethylzinc Alkoxides. *J. Am. Chem. Soc.* **2001**, *123*, 12160–12167.
- (14) Trimethylsulfoxonium- d_9 iodide was prepared from DMSO- d_6 (99.9%-D) and iodomethane- d_3 , prepared from methanol- d_4 (99.8%-D) and hydroiodic acid:²³ Gant, T. G.; Sarshar, S. Substituted phenethylamines. *US Pat.* 2008/0300316.
- (15) (a) Baldwin, J. E.; Bhatnagar, A. K.; Choi, S. C.; Shortridge, T. J. Rearrangement of Strained Dipolar Species, Aziridine N-Oxides. II. J. Am. Chem. Soc. 1971, 93, 4082–4084. (b) Albini, A. Synthetic Utility of Amine N-Oxides. Synthesis 1993, 263–277. (c) Penkett, C. S.; Simpson, I. D. Oxidative rearrangements of bicyclic 2-alkenyl aziridines. Tetrahedron Lett. 2001, 42, 3029–3032.
- (16) Recent reviews on hetero Diels-Alder reactions of nitroso compounds: (a) Yamamoto, H.; Kawasaki, M. Nitroso and Azo Compounds in Modern Organic Synthesis: Late Blooming but Very Rich. *Bull. Chem. Soc. Jpn.* 2007, 80, 595–607. (b) Carosso, S.; Miller, M. J. Nitroso Diels-Alder (NDA) reaction as an efficient tool for the functionalization of diene-containing natural products. *Org. Biomol. Chem.* 2014, 12, 7445–7468. (c) Brulíková, L.; Harrison, A.; Miller, M. J.; Hlaváč, J. Stereo- and regioselectivity of the hetero-Diels-Alder reaction of nitroso derivatives with conjugated dienes. *Beilstein J. Org. Chem.* 2016, 12, 1949–1980.
- (17) (a) Balasubramanian, M. 1,2-Oxazines and their Benzo Derivatives. in *Comprehensive Heterocyclic Chemistry III*, Vol. 8, Ramsden, C. A.; Scriven, E. F. V.; Taylor, R. J. K., Ed.; Elsevier: Oxford, 2008, pp. 333–371. (b) Utecht, G.; Jasiński, M. 3,6-Dihydro-2*H*-1,2-oxazines. *Chemistry of Heterocyclic Compounds* **2016**, *52*, 143–145.
- (18) Zheng, H.; McDonald, R.; Hall, D. G. Boronic Acid Catalysis for Mild and Selective [3+2] Dipolar Cycloadditions to Unsaturated Carboxylic Acids. *Chem. Eur. J.* **2010**, *16*, 5454–5460.
- (19) Chiacchio, U.; Liguori, A.; Romeo, G.; Sindona, G.; Uccella, N. Ring-Opening of Isoxazolidine System: Homologation of 3-Aryl into 3-Styryl Nitrones Via Intermediate 5-Hydroxy-Isoxazolidines. *Tetrahedron* 1992, 48, 9473–9482.
- (20) Unterhalt, B.; Eljabour, S. Ungesättigte Oxime, 29. Mitt. 1-Methoxyimino-3-pheny-propene.

- Arch. Pharm. 1986, 319, 1146-1150.
- (21) Hu, L.; Rombola, M.; Rawal, V. H. Synthesis of 1,2-Oxazinanes via Hydrogen Bond Mediated [3 + 3] Cycloaddition Reactions of Oxyallyl Cations with Nitrones. *Org. Lett.* **2018**, *20*, 5384–5388.
- (22) Carlsen, P. N.; Jiang, C.; Herrick, I. R.; Frontier, A. J. Studies toward the AB ring system of the tetrapetalone natural products. *Tetrahedron* **2015**, *71*, 5886–5896.
- (23) Cheng, Y.; Ding, W.-H.; Long, Q.; Zhao, M.; Yang, J.; Li, X.-Q. Synthesis of stable isotopically labelled 3-methylfuran-2(5*H*)-one and the corresponding strigolactones. *J. Label Compd. Radiopharm* **2015**, *58* 355–360.